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SILVEIRA, E.

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DENVER, CO

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13. ABSTRACT (Maximum 200 words)

METHODOLOGY AND RESULTS OF THE ANALYSIS OF 102 DENVER GROUND WATER SAMPLES FOR VARIOUS TARGET COMPOUNDS ARE PRESENTED. THE OBJECTIVE OF THE DENVER SOIL AND GROUND WATER PROJECT IS TO PREVENT FURTHER ENVIRONMENTAL CONTAMINATION, CONTROL/CONTAIN EXISTING CONTAMINATION, AND UNDERTAKE MITIGATIVE EFFORT AS DEEMED NECESSARY. AS PART OF THE PROBLEM DEFINITION PHASE OF THIS PROJECT A LARGE NUMBER OF DENVER WATER SAMPLES WERE COLLECTED AND ANALYZED AT BSRC. THE PURPOSE OF THIS SURVEY WAS THE CONFIRMATION OF SEMI-QUANTIFICATION OF SPECIFIC TARGET COMPOUNDS KNOWN OR SUSPECTED TO BE PRESENT IN THE WATER SAMPLES. GB/MS WAS THE METHOD OF CHOICE FOR THIS WORK; THE OVERALL METHODOLOGY BEING SIMILAR TO THAT EMPLOYED FOR PRIORITY POLLUTANT ANALYSIS OF WATER. ALSO DESIRED WAS IDENTIFICATION AND SEMIQUANTIFICATION OF ANY OTHER SIGNIFICANT COMPONENTS. IN ADDITION, PH, CONDUCTANCE, CHLORIDE ION, SULFATE ION, AND IONIC COPPER WERE DETERMINED.

DATA QUALITY INDICATED 1

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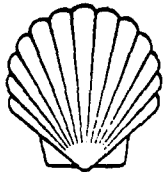
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SHELL CHEMICAL COMPANY

A DIVISION OF SHELL OIL COMPANY

P. O. BOX 2171

DENVER, COLORADO 80201

April 16, 1981

Colorado Department of Health
Water Quality Control Division
ATTN Mr. Robert Shukle, Industrial
Waste Consultant
4210 East 11th Avenue
Denver, CO 80220

Gentlemen:

Shell has conducted a survey of the groundwater on its leasehold on the Rocky Mountain Arsenal to assess groundwater quality. The analytical results from the survey as they became available have been reviewed with members of the Technical Review Committee at the last several meetings. These results have been summarized in a Technical Progress Report (M-9-81) which was recently issued by Shell Development Company. A copy of this report and a map showing well locations are attached for your information.

Yours very truly,

Original - 100-108976

J. H. Knaus
Plant Manager

Attachments

cc: Commander
Rocky Mountain Arsenal
Commerce City, CO 80022

Commander
Rocky Mountain Arsenal
ATTN SARRM-TOE-C, Greg Ward, Bldg. 741
Commerce City, CO 80022

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ANALYSIS OF DENVER WATER SAMPLES
TECHNICAL PROGRESS REPORT NO. M-9-81

By E. J. Silveira

ANALYSIS OF DENVER WATER SAMPLES



Project No. 83596.00

Denver Plant Ground Water Program

Technical Progress Report No. M-9-81

Date of Issue:

Written and Reviewed : E. J. Silveira
Participants : G. W. Campbell, D. R. Farinha,
K. S. Williams, B. Calog
Approval : R. A. Newman
References : MOLR 2383, 2606

SHARED - Under the Research Agreement between SIRM,
and Shell Oil Company dated January 1, 1960,
as amended.

ABSTRACT

Methodology and results of the analysis of 102 Denver groundwater, samples for various target compounds are presented.

LHINK Keywords: Water, analysis, Denver, groundwater, Rocky Mountain, arsenal, Colorado, VOA, pH, conductance, chloride, ion, sulfate, copper, method, run-off, well water, effluent

ANALYSIS OF DENVER WATER SAMPLES

Introduction

The objective of the Denver Soil and Groundwater Project is to prevent further environmental contamination, control/contain existing contamination, and undertake mitigative effort as deemed necessary. As part of the Problem Definition Phase of this project a large number of Denver water samples were collected and analyzed at BSRC. The purpose of this survey was the confirmation and semi-quantification of specific target compounds known or suspected to be present in the water samples. GC/MS was the method of choice for this work; the overall methodology being similar to that employed for priority pollutant analysis of water. The list of desired target compounds is shown in Table 1. Also desired was identification and semi-quantification of any other significant components.

In addition, pH, conductance, chloride ion, sulfate ion and ionic copper were determined.

Sampling and Storage

The sampling protocol for the Denver Plant Groundwater Survey Program is as follows.

Preparation of the Wells Before Sampling Water

Wells are to be drilled at designated spots on the properties, lined with polyvinyl chloride pipe, and capped. Each well is to be numbered by stencil "Denver Plant Groundwater Well No. ____" (1 through n). Each well is to have an altitude benchmark referenced to the U.S.G.S. benchmark. Prior to each sampling of the well water, the height of the water will be determined, measured from the top of the well, and referenced to the benchmark. By use of a submersible pump (pumps which require priming or vacuum are unsuitable) and a length of Teflon tubing, about 10 volumes of water (as compared to the amount in the well casing) will be pumped out of the well. If recovery times are short and the pumping operation can be accomplished in a reasonable time frame, pumping will be done before each sampling. After each pumping operation, Denver drinking water will be used to rinse the pump. The Teflon tube should require no rinsing but may require drying by use of a piece of gauze or similar clean material.

Sampling the Water

In advance of the sampling operation, new quart bottles will be rinsed with methylene chloride and dried at 100°C. Smaller bottles or vials, 40-ml, will be supplied by Westhollow Research Center (WRC)(G. W. Stanko). All bottle caps will contain Teflon liners also to be supplied by WRC. Each bottle will be carefully stored and handled so as to avoid contamination.

The actual water sample will be obtained by bailing as follows: Immediately after pumping, the sample bottle will be placed into a weighted, stainless cage. By use of a stainless steel chain, the bottle will be quickly lowered to the bottom of the well and allowed to fill with water. The full bottle will be brought to the surface, capped, labeled with tape labels, and placed in an

ice bath. The 40-ml size bottle will be sampled in a similar manner, with care being taken to insure liquid full bottles (no air space) upon insertion of the Teflon septum. Four of the 40-ml bottles will be placed in a wide-mouth quart jar and cushioned with wads of paper. After capping, the quart jar will be placed in an ice bath. Two quart and three 40-ml samples of water shall be obtained from each well at each sampling. The full bottles will be returned to the laboratory and placed in a clean refrigerator at a temperature of 0 to 5°C (32 to 41°F); freezing must be avoided. The refrigerator will be used for no other purpose during this groundwater survey program. After the completion of the sampling for the period, or when about 20 wells have been sampled, the samples will be sent to Modesto (BSRC) by refrigerated truck or by air freight in a manner which will keep the bottles between 0 and 5°C. Upon receipt of the samples at BSRC, they will be placed in a refrigerator (0 to 5°C) pending analysis.^{a)}

Sampling of the wells at Denver will be coordinated with the work load at BSRC to insure volatile analyses within two weeks of sampling. Extraction should be accomplished within one to two weeks of sampling. Extracts may be sealed in airtight vials and stored in a freezer to await analysis.

Records

A log book will be maintained for recording all sampling data. Among the entries will be the names of persons doing the sampling, weather data, height of the water in the well, any variance in the sampling procedure, and unusual events. The log book will be signed by the responsible person after each series of sampling.

A map showing the location of each well is available on request from the Denver Plant.

Instrumentation and Methodology for Volatile Organics Analysis (VOA)

The general scheme for VOA analysis is shown in flowchart form in Figure 1. Specific analytical conditions are given in Table 2.

VOA Subsampling Procedure

The VOA vial was removed from the refrigerator and allowed to come up to room temperature. The vial was opened and the tip of a 1-ml syringe (no needle attached) was inserted well below the surface and a 1.0-ml sample was drawn very slowly into the syringe. The internal standard was then added by syringe directly into the subsample. A needle was then immediately attached and the sample was injected into the sparging vessel through the septum valve.

VOA Standards

VOA reference standard solutions were freshly prepared at least once each week. The reference standard contained 20 µg/ml each of the internal standard and the target compounds in methanol solution. At least one reference standard run was made each day that samples were run. The reference standard run sample was prepared by syringe addition of 5 µl of reference standard solution to 1.0-ml of distilled water previously boiled to remove volatiles. The purpose of this reference standard run was to (1) ensure the reliability of the analysis, and (2) to update response

a) The procedure for analysis is described separately.

factors upon which the quantification was based. A typical VOA reference standard run is shown in Figure 2.

VOA internal standard solutions were freshly prepared at least once each week. This solution contained 20 $\mu\text{g}/\text{ml}$ of only the internal standard, bromochloromethane, in methanol. A 5 μl aliquot of the internal standard solution (100 ng bromochloromethane) was added by syringe to each sample and blank immediately prior to introduction to the sparging vessel. Blank runs, containing only internal standard added to distilled, previously boiled water, were made at least once each day.

VOA reference standard and internal standard solutions were kept refrigerated when not in use.

A typical gas chromatogram/mass spectrum (GC/MS) of the VOA is shown in Figure 2.

Extraction Procedure

Filtration - Prior to extraction samples were filtered through a 0.45 μm Millipore membrane filter, catalogue number HAWP-047-00, covered with a Millipore glass fiber prefilter, catalogue number AP25-047-00. A Millipore filter holder with glass frit was used.

pH Adjustment

Prior to pH adjustment and extraction approximately 100-ml of each sample was retained and submitted for pH, conductance, chloride ion, sulfate ion and ionic copper determinations.

The volume of the remaining filtrate was noted and the filtrate transferred to a 2-l separatory funnel. Well samples 1A-46A were subjected to extraction without pH adjustment.

The pH of all other samples were adjusted to pH 11 with sodium hydroxide pellets. After extraction the aqueous phase was then adjusted to pH 2 with 6N HCl and reextracted.

Extraction

The aqueous solutions were extracted with 120-ml CH_2Cl_2 and the CH_2Cl_2 phase collected. The aqueous solutions were then extracted twice more with 80-ml portions of fresh CH_2Cl_2 and the three extracts combined.

Concentration

The combined CH_2Cl_2 extracts were concentrated as rapidly as possible in a Kuderna-Danish Evaporator equipped with a 3-ball Snyder column to a volume of approximately 10-ml. The extracts were then further concentrated to approximately 0.8-ml by blowdown with dry nitrogen. An aliquot of a standard solution of d_{10} -anthracene internal standard (containing 100 μg) was then added to the extract concentrates and the volume adjusted to 1.0-ml. The extracts derived from the basic solutions are the base-neutral (BN) extracts and those derived from the acidic solutions are the acid extractables (AE). Those extracts derived from solutions without pH adjustment are listed in the result table under the BN heading.

Instrumentation and Methodology for GC/MS Analysis for Extractable Organics

The general scheme for extractable organics analysis is shown in flowchart form in Figure 3. Specific analytical conditions are given in Table 3.

BN Standards

BN reference standard solutions were freshly prepared at least every 3-4 weeks. The reference standard contained 100 µg/ml each of the internal standard and the target compounds in methylene chloride solution. At least one reference standard run was made each day that samples were run. The purpose of this reference standard run was (1) to ensure the reliability of the analysis, and (2) to update response factors upon which the quantification was based. A typical reference standard run is shown in Figure 4.

BN internal standard solutions were freshly prepared at least every 3-4 weeks. This solution contained 1 mg/ml of only the internal standard, d₁₀-anthracene, in methylene chloride. A 100 µl aliquot of the internal standard solution (100 µg d₁₀-anthracene) was added by syringe to each sample extract after nitrogen blowdown.

BN reference standard and internal standard solutions were kept refrigerated when not in use.

A typical GC/MS of a BN extract is shown in Figure 4.

Samples 1A-4A, 9A, 13A, 16A, 17A, 23A-25A, 32A-34A, and 39A-41A were run on the OV-101 packed column (conditions A in Table 3). Samples 5A-8A, 10A-12A, 14A, 15A, 18A-22A, 26A-31A, 35A-38A, 42A-46A, 16B, 24B, 40B, were run on the glass capillary column (conditions B in Table 3). All other samples were run on the fused silica capillary column (conditions C in Table 3).

Data Processing of GC/MS Runs

The general scheme for data processing is shown in flowchart form in Figure 5. Details of the data processing procedures and programs are to be documented in a separate report.

pH Measurement

The pH of the water samples was determined by using a Corning 110 Digital pH Meter with a Metrohm EA 121 combination glass electrode.

Conductance Measurement

Conductance was determined using a Western Instruments Model 210 Conductivity Meter with a Model 219-200 flow through cell. The calibration solution was 0.01 M KCl.

Chloride Ion Determination

Chloride ion was determined by silver nitrate titration using a Metrohm E 436 Potentiograph with a Metrohm EA 246 silver electrode.

Sulfate Ion Determination

Sulfate ion was determined turbidimetrically as BaSO_4 using a Hach Model 2100 Turbidimeter.

Ionic Copper Determination

Ionic copper was determined using a colorimetric method - EMS 2C14/59.

Results

The analytical results are presented in Table 4. Concentration values (in $\mu\text{g}/\text{l}$ or ppb) are given for target compounds exceeding the nominal minimum detectable concentration of 10 ppb. Concentration values for other detected peaks, including tri- and tetrachlorethylene, DMK (acetone), CH_2Cl_2 and THF, are given if the peak could be tentatively identified and the concentration exceeded 200 ppb.

FIGURE 1

VOA SCHEME

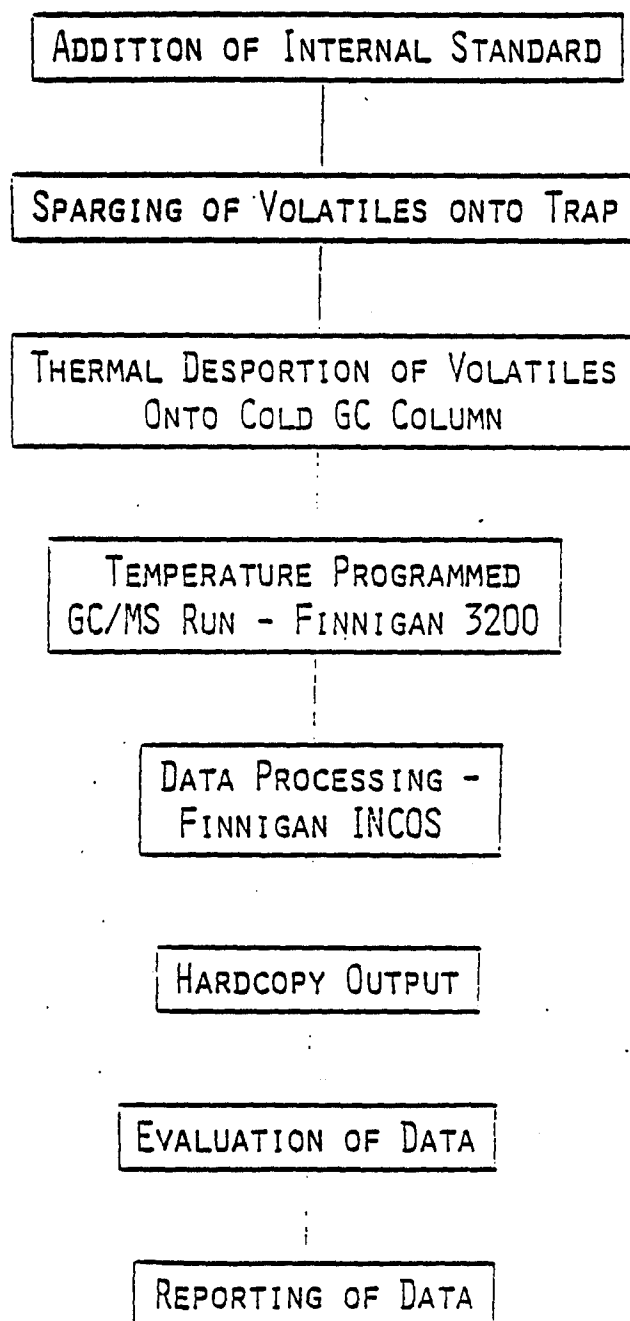


FIGURE 2
VOA STANDARDS

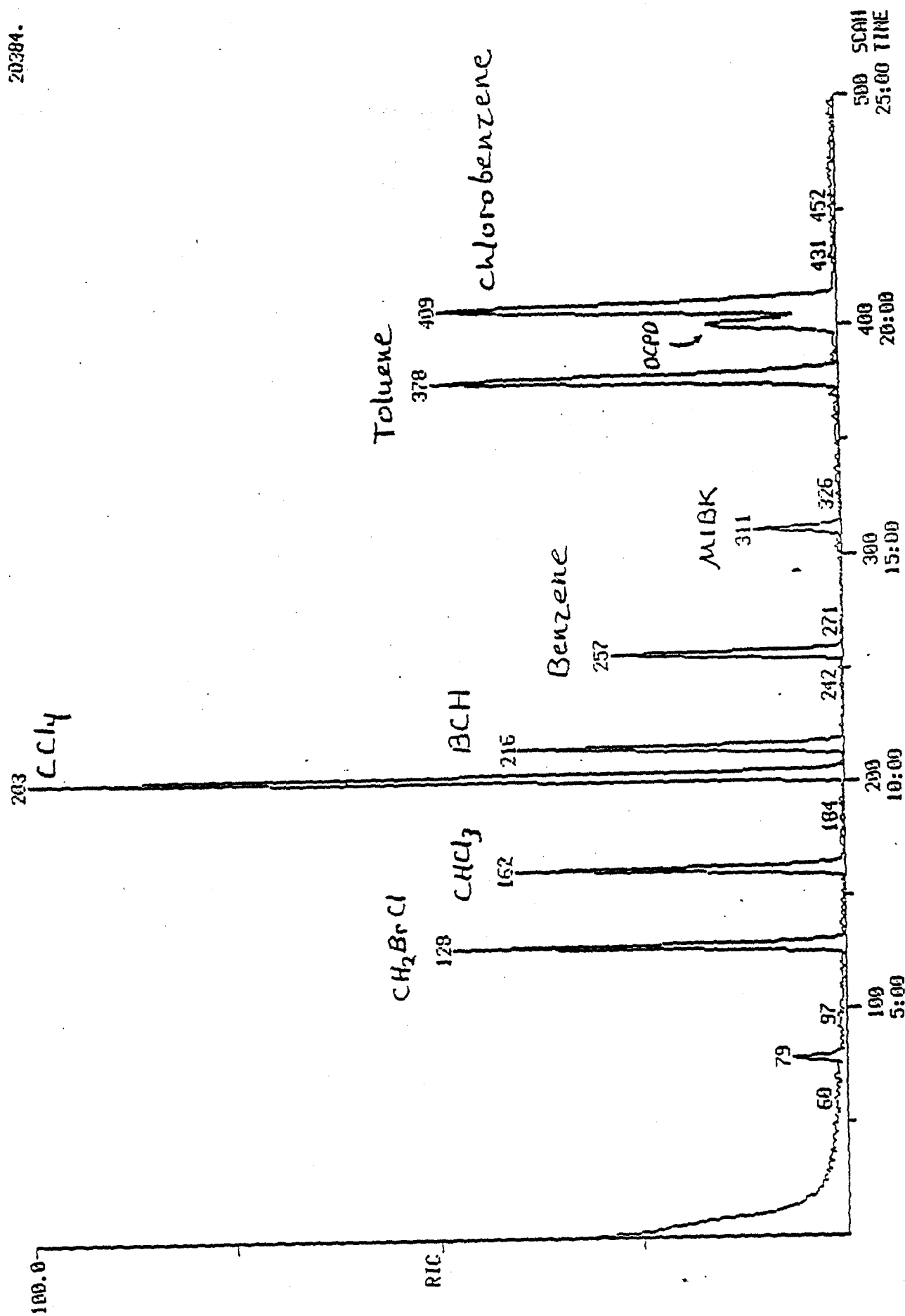
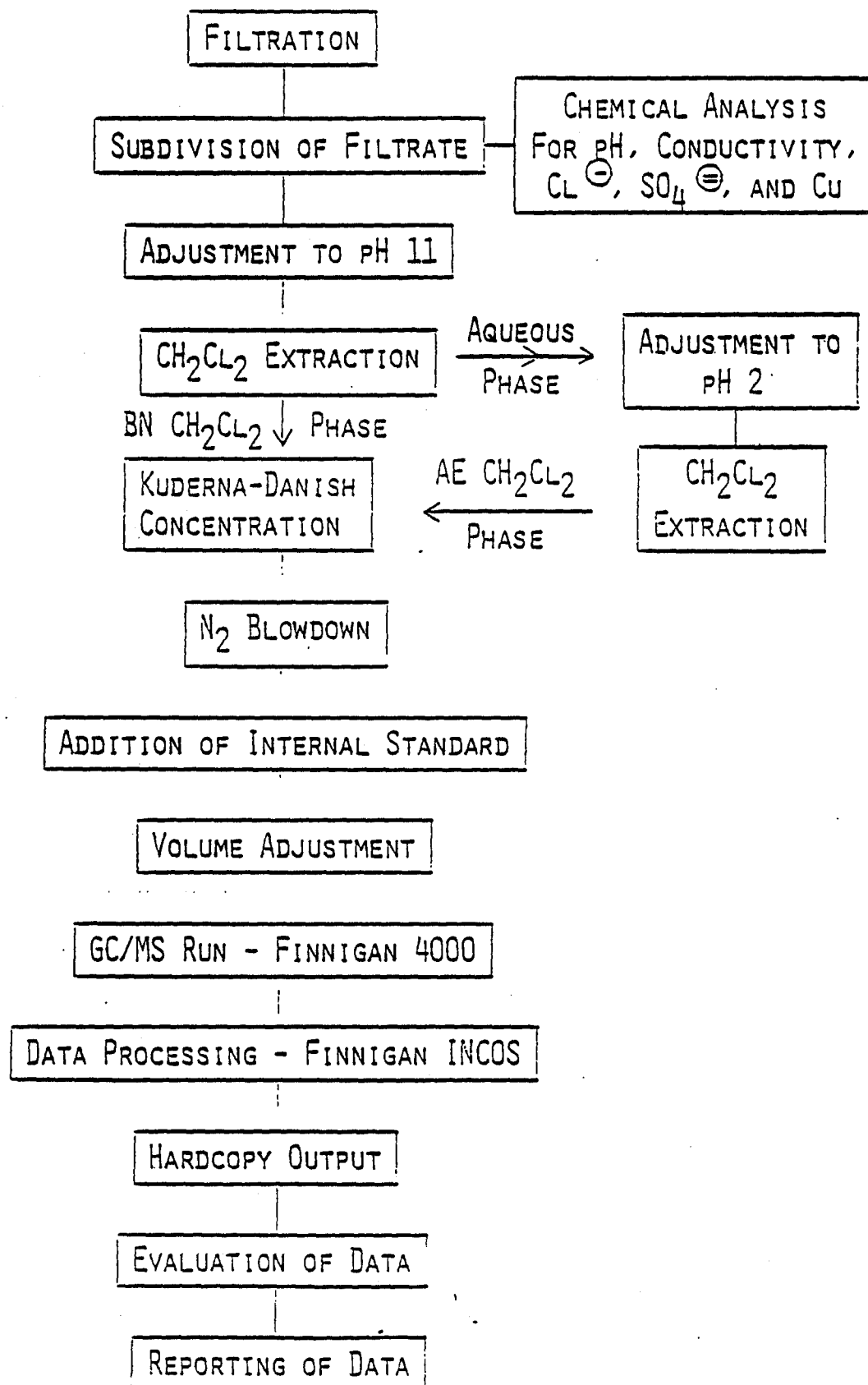


FIGURE 3

LLE SCHEME



FIGURE

BN STANDARDS

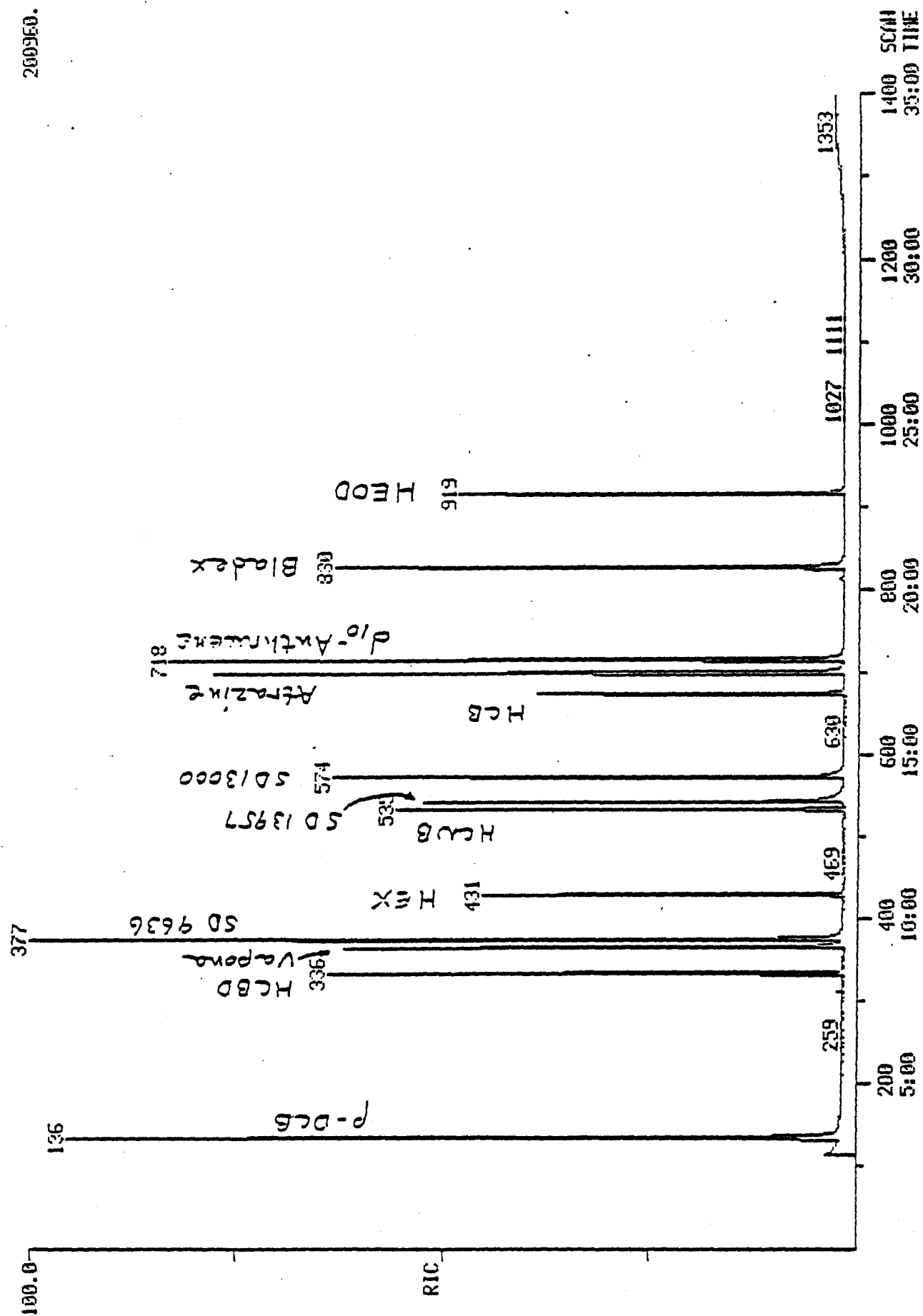
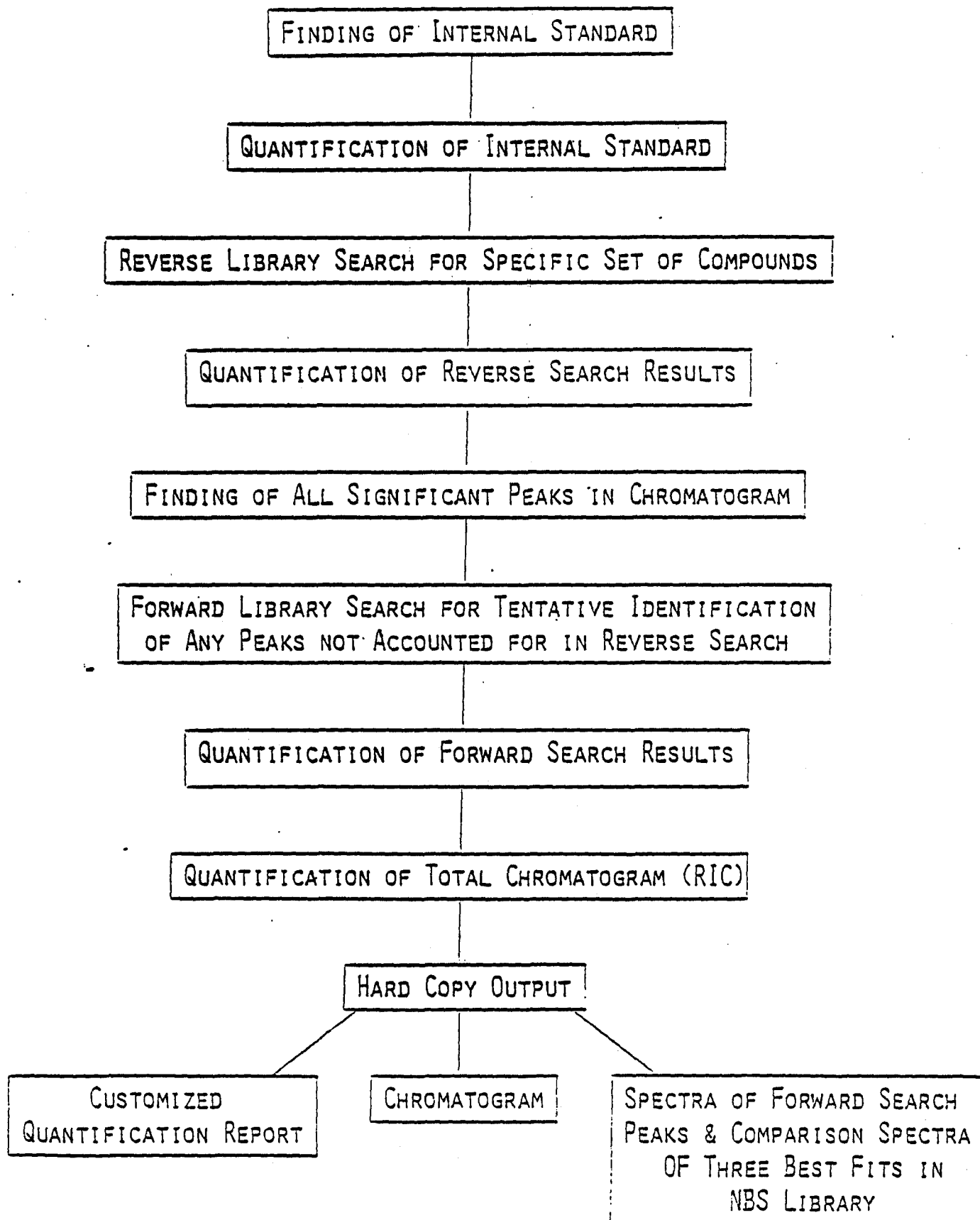





FIGURE 5


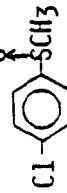
DATA PROCESSING SCHEME



DENVER PLANT GROUNDWATER ANALYSIS

VOLATILE ORGANICS	MW	QUANTIFICATION ION	ASSIGNMENT	RELATIVE ABUNDANCE
1. CHLOROFORM	118	83	M ⁺ - CL	100
2. CARBON TETRACHLORIDE	152	117	M ⁺ - CL	100
3. BICYCLOHEPTADIENE (BCH)	92	91	M ⁺ - H	100
				
4. BENZENE	78	78	M ⁺	100
- 5. METHYLISOBUTYLKETONE (MIBK)	100	100	M ⁺	
6. TOLUENE	92	91	M ⁺ - H	100
7. DICYCLOPENTADIENE (DCPD)	132	66		100
				
8. CHLOROBENZENE	112	112	M ⁺	100

DENVER PLANT GROUNDWATER ANALYSIS

EXTRACTABLE ORGANICS	MW	QUANTIFICATION ION	ASSIGNMENT	RELATIVE ABUNDANCE
1. DICHLOOROBENZENES	146	146	M ⁺	100
2. DIISOPROPYLMETHYLPHOSPHONATE (DIMP)	180	97	$\begin{array}{c} \text{HO}-\text{P}-\text{OH}^+ \\ \quad \\ \text{HO} \quad \text{CH}_3 \end{array}$	100
$\begin{array}{c} (\text{CH}_3)_2\text{CHO}-\text{P}-\text{O} \\ \quad \\ (\text{CH}_3)_2\text{CHO} \quad \text{CH}_3 \end{array}$				
3. DIBROMOCHLOROPROPANE (DBCP)	234	157	M ⁺ - Br (P+2 ISOTOPE)	100
CH ₂ BrCHBrCH ₂ Cl				
4. HEXACHLOROBUTADIENE	258	225	M ⁺ - Cl (P+2 ISOTOPE)	100
5. VAPONA	220	109	(CH ₃ O) ₂ P=O ⁺	
$\begin{array}{c} \text{CH}_3\text{O}-\text{P}-\text{O} \\ \quad \\ \text{CH}_3\text{O} \quad \text{OCH=CCl}_2 \end{array}$				
6. SD 9636	158	158	M ⁺	100
				
7. HEXACHLOROCYCLOPENTADIENE (HCEX)	270	237	M ⁺ - Cl (P+2 ISOTOPE)	
8. HEXACHLORONORBORBIADIENE (HCNB)	296	263	M ⁺ - Cl (P+2 ISOTOPE)	100
9. SD 13957	174	159	M ⁺ - CH ₃	100
				

RELATIVE ABUNDANCE

ASSIGNMENT

QUANTIFICATION ION

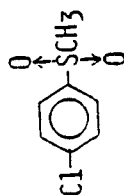
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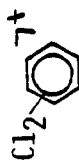


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10. SD 13000



100



142

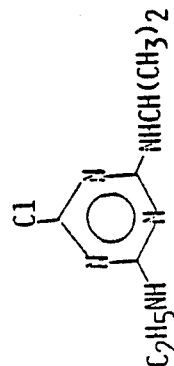
11. HEXACHLOROBENZENE (HCB)

65

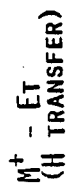


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12. ATRAZINE

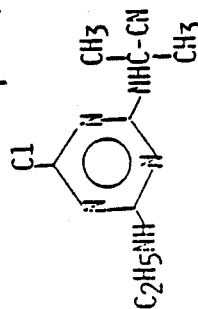


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212

13. BLADEX



100



79

14. DIELDRIN (HEOD)



OTHER COMPOUNDS WHICH HAVE BEEN DETERMINED IN SELECTED SAMPLES BY
SPECIFIC RESIDUE METHODS:

1. TRIMETHYL PHOSPHATE
2. SD 15042

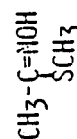


TABLE 2 - Conditions for Volatile Organics Analysis

Instrument: Finnigan 3200 GC/MS equipped with a Chemical Data Systems
Model 310 Concentrator

Trapping Column: 2' x 1/8" OD stainless steel packed with Tenax 60/80 mesh

Sparge Gas: Helium

Sparge Rate: 30 ml/minute

Sparge Cycle: 12 minutes with sparging vessel at 50°C onto trapping
column at room temperature

Desorb Cycle: 4 minutes at 180°C onto GC column at room temperature

Trap Bakeout Cycle: 4 minutes at 280°C

Sample Size: 1 ml or 5 ml

GC Column: 10' x 2mm ID glass column packed with 1% SP-1000 on
Carbopack B, 60/80 mesh

Temperature Program: 50°C for 4 minutes, then programmed to 200°C
at 10°/minute, then held at 200°C for 13.5 minutes

Injector Temperature: 200°C

Carrier: Helium at 30 ml/minute

Separator: Glass jet

Separator Temperature: 200°C

Transfer Line Temperature: 200°C

MS Mode: Electron Impact (EI)

Electron Energy: 70 eV

Electron Multiplier Voltage: Approximately 1700 V

Emission Current: 1.0 ma

Preamplifier Sensitivity: 10^{-7} a/v

Scan Range: 40-540 amu

Scan Time: 3 seconds

Scans per Run: 650

Internal Standard: 100 ng bromochloromethane

TABLE 3 - Conditions for Extractable Organics Analysis

Instrument: Finnigan 4000 GC/MS

GC Columns: (A) 6' x 2mm ID glass column packed with 3% OV-101 on Gas Chrom Q, 80/100 mesh
(B) 30m x 0.25mm ID glass WCOT capillary column coated with SE-54
(C) 30m x 0.25mm ID fused silica WCOT capillary column coated with SE-54

Temperature Program: (A) 50°C for 4 minutes, then programmed to 260°C at 10°/minute
(B) 70°C for 4 minutes, then programmed to 280°C at 8°/minute
(C) 70°C for 4 minutes, then programmed to 300°C at 8°/minute

Injector Temperature: 280°C

Carrier Gas: Helium

Flow Rate: 20 ml/minute (A)

Injector: Modified German-Horning (B and C)

Injector Flow: 20 ml/minute (B and C)

Injector Pressure: 14 psi (B and C)

Split Ratio: Approximately 10:1 (B and C)

Separator: Glass jet (A)

Separator Oven Temperature: 280°C

Transfer Line Temperature: 280°C

MS Mode: Electron Impact (EI)

Electron Energy: 70 eV

Electron Multiplier Voltage: Approximately 1700 V (without conversion dynode) or approximately 1000 V (with conversion dynode at 3000 V)

Emission Current: 0.2 ma

Preamplifier Sensitivity: 10^{-7} a/v

Scan Range: 40-640 amu

Scan Time: (A) 3 seconds; (B) 2 seconds; (C) 1.5 seconds

Scan per Run: (A) 500; (B) 900; (C) 1400

Volume Injected: 2 or 3 μ l

Internal Standard: 100 μ g/ml d_{10} -anthracene in final extract

TABLE 4

Water Sample	CHCl ₃	CCl ₄	BCH	Benzene	MIBK	Toluene	DCPD	Chlorobenzene	dichloro benzenes	DIMP	LACP	HCSD	Tapona	SD 9636	HEX	HCNB	SD 13957	SD 13000	HCB	Atrazine
1A				140																
2A	2900 ⁽¹⁾		84	170	1100	44	370	110										17		
3A	2000	230	9500 ⁽¹⁾	2300 ⁽¹⁾	130	370	3100	5200	660											
4A	470		8000 ⁽¹⁾	3000 ⁽¹⁾		2600	15000 ⁽¹⁾		33											
5A	18000 ⁽¹⁾	160		1700 ⁽¹⁾	15			850	150 ⁽⁶⁵⁾		27 ⁽⁶⁵⁾							68 ⁽⁶⁵⁾		
6A	4200 ⁽¹⁾		88	180	1200	48	400	120	30									27 ⁽⁶⁵⁾		
7A	740			26														17 ⁽⁶⁴⁾		
8A	4700			100				11												
9A	2900			55																
10A	1200	1500		370				230	85		47									
11A	12			23		15												10		
12A	250	35		42				34												
13A	20000 ⁽¹⁾		31	1900 ⁽¹⁾	1700	2600	560	70						170				1700		
14A	1100	1200		490				380	23 ⁽⁶⁵⁾					33 ⁽⁶⁵⁾				120 ⁽⁶⁵⁾		
15A	1000 ⁽¹⁾	33		140			720 ⁽¹⁾	76	15 ⁽⁶⁵⁾		20 ⁽⁶⁴⁾							11 ⁽⁶⁵⁾		
16A	170	19		110														41		
17A	3600 ⁽¹⁾	1800 ⁽¹⁾	1000 ⁽¹⁾	1000 ⁽¹⁾	37	740 ⁽¹⁾		1700 ⁽¹⁾	51		7100 ⁽¹⁾	18						60		
18A	2900 ⁽¹⁾	750		49				29	500 ⁽¹⁶⁵⁾		120 ⁽⁶⁵⁾							25 ⁽⁶⁵⁾		
19A	2900 ⁽¹⁾	1200	450	24		74					190							660		
20A			2600	760				49	48					17				320		59
21A	11			5800 ⁽¹⁾				35												
22A	30			12		120	26											150		
23A	240	25		36							25									
24A	250	74	74	1200 ⁽¹⁾	4400	1200 ⁽¹⁾		2100 ⁽¹⁾	3400 ⁽¹⁾					89			17	22		
25A	1500 ⁽¹⁾			280	650	560	120	210						730				7300		
26A	9800 ⁽¹⁾	2500	310	930		3700 ⁽¹⁾					30000 ⁽¹⁾							96		
27A	6000 ⁽¹⁾	1200	5000 ⁽¹⁾	2700 ⁽¹⁾		3400 ⁽¹⁾		420	670		560 ⁽⁶⁴⁾							73 ⁽⁶⁴⁾		
28A				21			60											61		
29A	19			5800 ⁽¹⁾			31											20		
30A	24			3900 ⁽¹⁾																
31A																				
32A	55			8700 ⁽¹⁾			97													
33A	86		260	21000 ⁽¹⁾		630	2700 ⁽¹⁾													

All concentrations are in µg/l (ppb)

Water Sample	Bladex	HEAD	trichloroethylene	1,1,1-trichloroethylene	DMK	CH ₂ Cl ₂	THF	Others	VOA	BT	AP	pH	Conductance Ohm-cm x 10 ⁻³	Chloride mg/L	Sulfate mg/L	Copper ppm	SD 15062-94	TUP095
1A		280	660									8.2	0.55	0.033	0.036	<0.5		
2A	110	490					2) 3)					7.6	0.96	0.081	0.13	<0.5		<1
3A		210	630	280			4)					11.9	3.0	0.18	0.17	<0.5		1
4A		990	6800 ¹⁾	450	500	5) 6) 7)						12.4	10.0	0.72	0.24	<0.5		<1
5A		520	7600 ¹⁾	230								7.4	3.0	0.22	0.20	<0.5		2
6A							8) 9)					7.6	4.5	1.4	0.62	<0.5		100
7A												8.1	0.37	<0.02	0.033	<0.5		<1
8A		480										7.3	3.8	1.1	0.36	<0.5 (trace)		<1
9A												7.5	1.4	0.076	0.12	<0.5		<1
10A					210							8.6	0.82	0.073	0.10	<0.5		
11A					680							8.2	1.0	0.065	0.18	<0.5		
12A												8.1	2.3	0.37	0.40	<0.5 (trace)		
13A		72	260	4900	1400	2700	10) 11)					9.2	3.0	0.80	0.17	<0.5	205 (8)	190
14A			430	6400			12)					7.8	1.5	0.19	0.12	<0.5	<10 (8)	
15A												7.4	2.3	0.61	0.15	<0.5	1400 (8)	
16A	220		21000									8.0	1.1	0.15	0.11	<0.5	15 (8)	
17A			2400	1100	2200 ¹⁾		13) 14) 15)					7.2	1.8	0.28	0.24	<0.5		70
18A						200	16)					7.2	1.4	0.27	0.12	<0.5		
19A							62)					7.8	1.6	0.18	0.30	<0.5		<1
20A	250	13										7.2	0.96	0.12	0.10	<0.5		
21A						330	41)					8.1	1.3	0.14	0.23	<0.5		
22A												8.3	1.3	0.18	0.24	<0.5		
23A												8.1	0.65	0.020	0.064	<0.5		
24A		790	340	2700 ¹⁾			17) 18) 19) 63)					8.0	1.0	0.17	0.10	<0.5		
25A			230	980 ¹⁾	300		20) 21) 22)					7.3	14.0	1.9	0.72	<0.5		140
26A				3200	690		36) 37) 38) 39) 40)					7.5	2.3	0.60	0.11	<0.5		
27A			510	400								7.3	1.4	0.28	0.24	<0.5		80
28A							78)					7.6	2.8	0.033	0.42	<0.5		
29A												8.3	0.82	0.054	0.083	<0.5		
30A						240						8.4	0.96	0.14	0.072	<0.5		
31A						220						7.9	1.0	0.056	0.26	<0.5		
32A							23) 24) 25) 26) 27)					8.1	0.96	0.020	0.14	<0.5		
33A							79) 80) 81) 82) 83)					7.8	1.1	0.043	0.26	<0.5		

All concentrations are in mg/l (ppb)

All concentrations are in $\mu\text{g/l}$ (μppb)

Water Sample	Bioindex	HEAD	Trichloroethylene	Trichloroethylene	DYK	CH ₂ Cl ₂	THF	Others	VOC	EW	MLG Total	pH	Conductance Ohm-cm-x 10 ⁻³	Chloride mg/ml	Sulfate mg/ml	Copper ppm	SD 15042544	TUP053
34A								28) 29) 30) 31) 32)				7.9	1.4	0.19	0.20	<0.5		
35A												8.0	1.4	0.17	0.25	<0.5		
36A												8.3	1.7	0.17	0.42	<0.5		
37A												8.4	0.58	<0.01	0.035	<0.5		
38A												7.6	3.2	0.074	1.2	<0.5		
39A								33)				8.0	2.2	0.60	0.20	<0.5		
40A								34)				7.6	2.6	0.20	0.72	<0.5		
41A								35)				8.1	2.3	0.65	0.17	<0.5		
42A												8.0	2.5	0.30	0.76	<0.5		
43A												8.4	0.69	0.043	0.072	<0.5		
44A												7.6	2.2	0.65	0.11	<0.5		
45A												8.0	0.76	0.066	0.086	<0.5		
46A												7.7	0.42	0.048	0.069	<0.5		
47A												7.6	1.7	0.26	0.43	<0.5		
48A												7.7	0.85	0.056	0.12	<0.5		
49A												7.9	0.98	0.074	0.15	<0.5		
50A												8.0	0.88	0.068	0.11	<0.5		
51A												7.4	3.3	0.083	0.68	<0.5		
52A												7.3	1.2	0.14	0.19	<0.5		
53A																		
54A												7.0	1.8	0.16	0.57	<0.5		
55A												7.7	2.4	0.62	0.22	<0.5		
56A												7.2	2.8	0.86	0.18	<0.5		
57A												7.2	2.6	0.78	0.16	<0.5		
58A												7.5	2.6	0.77	0.19	<0.5		
59A												7.7	2.6	0.77	0.19	<0.5		
60A												7.7	1.0	0.12	0.16	<0.5		
61A												7.7	1.8	0.38	0.23	<0.5		
62A												7.2	2.2	0.25	0.82	<0.5		
63A												7.5	1.2	0.13	0.19	<0.5		
64A												7.8	1.4	0.13	0.26	<0.5		
65A												7.4	1.3	0.12	0.27	<0.5		

All concentrations are in ug/l (ppb)

Water Sample	CHCl3	CCl4	BCH	Benzene	MIBK	Toluene	DCPD	Chlorobenzene	Dichlorobenzene	DIMP	BSCP	HCBP	Hex	HCVB	SD 13957	SD 13000	HCB	Atrazine
72A							110	15								110		
73A				500	94	26	76	68										
74A					870			83	11									
75A	38			1500	1700			270	16									
76A					300													
77A	440				220													
78A	12																	
79A				43	3700			1200	51								12	
80A							18										41	
81A					70													
82A	230	20		3600 ¹⁾				130	15									
83A	670	20		15000	35			270	15									
84A				3900	470													
85A																		
86A																		
87A																		
88A				680	3600 ¹⁾	210	17	75							68	12		
89A				2600 ¹⁾	5400 ¹⁾	370	3600 ¹⁾	2500										
90A	67	420		36	1500													
91A	6200 ¹⁾	4900 ¹⁾		100	450	530									180	1400		
92A	1900	6300 ¹⁾		600	690	18		16							32			
93A	120	280			100										2900	41		
94A	380	33					320								310	40		
16B	270	19		48												34	19	
24B	440	180		180	2100 ¹⁾	13000 ¹⁾	1900 ¹⁾	7200 ¹⁾	29							11		
27B				180			2200 ¹⁾											
38B				13	1800 ¹⁾		13	1200										
40B					13000 ¹⁾		790 ¹⁾	170	21000 ¹⁾									
52B				2100 ¹⁾	12000 ¹⁾	11	7500 ¹⁾	2600							76	15	26	
55B					2100 ¹⁾		600	58										
58B				160	4000 ¹⁾		3800 ¹⁾	270										

All concentrations are in µg/l (ppb)

Water Sample	Bladex	HEOD	trichloro ethylene	tetrachloro ethylene	DMK	CH ₂ Cl ₂	THP	Others	VOA	BN	AE	PH	Conductance Osm-Cu-X 10-3	Chloride mg/ml	Sulfate mg/ml	Copper ppm	SD 15402 96)	TUPO 95)
72A						660			760	60	84	7.3	1.5	0.21	0.17	<0.5		
73A									1600	110	170	7.2	1.5	0.18	0.16	<0.5		
74A						640			2000	250	76	7.3	1.7	0.24	0.18	<0.5		
75A						1100			4500	170	230	7.0	4.1	1.5	0.39	<0.5		
76A						200			790	79	73	7.5	0.96	1.4	0.10	<0.5		
77A						420			1300	120	150	7.1	4.5	1.8	0.10	<0.5		
78A									640	77	92	7.4	3.6	1.2	0.23	<0.5		
79A						2600			7200	360	51	7.3	3.1	0.94	0.27	<0.5		
80A									1100	220	60	7.7	1.4	0.15	0.15	<0.5		
81A									790	210	85	7.4	1.3	0.13	0.15	<0.5		
82A									2300	79	63	7.3	1.7	0.21	0.31	<0.5		
83A									29000	670	260	7.1	2.9	0.62	0.75	<0.5		
84A						360			5400	94	91	7.6	1.1	0.16	0.17	<0.5		
85A						230			860	76	75	7.8	0.81	0.08	0.08	<0.5		
86A									86	110	58	7.5	0.83	0.08	0.07	<0.5		
87A									90	33	110	7.8	0.90	0.10	0.10	<0.5		
88A						530	84)		19000	1200	220	7.6	1.0	0.14	0.09	<0.5		
89A							61)		19000	25000	3400	8.0	0.64	0.06	0.07	<0.5		
90A						660			4000	330	100	7.6	2.2	0.44	0.43	<0.5		
91A						2300 1)	89)		18000	8800	3000	7.4	2.1	0.40	0.43	<0.5		
92A									11000	260	100	7.7	1.9	0.29	0.37	<0.5		
93A	27					690			1500	730	190	7.7	2.1	0.36	0.39	<0.5		
94A	350								14000	930	38	7.1	7.0	3.3	0.25	<0.5		
16B	260	20000 1)							970	1100	7.9	1.3	0.24	0.09	<0.5			
24B		2000	7400 1)				68) 69)		470	760	7.7	0.99	0.19	0.09	<0.5			
37B									38000	10000	7.8	0.64	<0.01	0.03	<0.5			
38B									3500	980	2700	7.0	3.6	0.07	1.2	<0.5		
40B									26000	1000	1000	7.7	3.3	0.40	0.74	<0.5		
52B									55000	170000	35000	7.8	1.1	0.13	0.16	<0.5		
55B						200	67)		55000	2500	760	7.4	2.5	0.65	0.19	<0.5		
58B							66)		100000	4600	1700	7.2	2.5	0.78	0.15	<0.5		

All concentrations are in ug/l (ppb)

Water Sample	CHCl ₃	CCl ₄	BCH	Benzene	MIBK	Toluene	DCPD	Chlorobenzene	Dichloro Benzenes	UHP	DACP	HCB	SD 3036	HEX	HCNB	SD 13957	SD 13190	HCB	Atrazine
59B	12	13000	1	270	89														14
61B		9400	1																16
65B	190	3700	1	180	32														600
77B	810	520																	540
78B	23		44																
79B			10																
80B	21		38																520

Water Sample	Bladex	HEOD	Trichloro ethylene	Trichloro ethylene	1,1,1-Trichloro ethane	1,1,2-Trichloro ethane	CH ₂ Cl ₂	UHP	Substrates	VOA	EW	RIE Total	PH	Conductance (µm-Cu-X-10-3)	Chloride mg/ml	Sulfate mg/ml	Copper ppm	TUPO 95	SD 15402 94
59B										73000	1300	820	7.7	2.5	0.78	0.16	<0.5		
61B										60000	150	230	7.3	1.8	0.55	0.18	<0.5		
65B										210000	30000	14000	93	93	93	93	93		
77B								230		1200	550	220	7.3	4.2	1.7	0.10	<0.5		
78B										450	450	300	7.6	2.7	0.91	0.15	<0.5		
79B										1800	260	250	7.5	2.8	0.85	0.24	<0.5		
80B										750	610	200	7.7	1.3	0.16	0.13	<0.5		

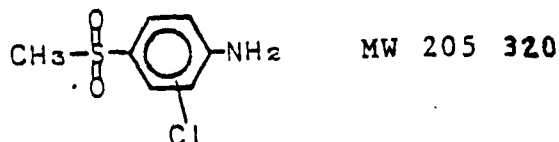
All concentrations are in µg/l (ppb)

Footnotes to Table 4

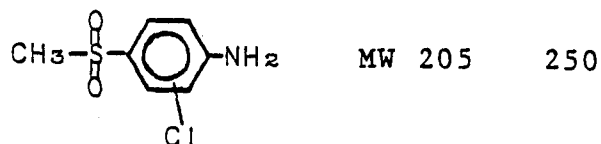
- 1) At least one ion in spectrum saturated; actual concentration probably greater.
- 2) Dimethyldisulfide 3600¹⁾
- 3) 1,2-bis(methylthio)ethane 560
- 4) Dichloroethene 280
- 5) Cyclopentene 1900
- 6) Cyclopentadiene 400
- 7) Methyl cyclohexane 500
- 8) Dimethyldisulfide 4200¹⁾
- 9) 1,2-bis(methylthio)ethane 590
- 10) Methyl cyclohexane 250
- 11) Ethyl benzene 4500
- 12) Hexachloroethane 1800
- 13) Dichloropropene 210
- 14) Bromochloropropene 310
- 15) Ethyl benzene 630
- 16) Bromodichloro methane 520
- 17) 2-propanol 360
- 18) Dichloroethene 300
- 19) Methyl cyclohexane 300
- 20) Ethyl benzene 340
- 21) Methoxy benzene 440
- 22) Dimethyl disulfido 1200¹⁾
- 23) Cyclopentene 360
- 24) Cyclohexane 260
- 25) Hexane 400
- 26) Methyl cyclohexane 450
- 27) Ethyl benzene 490
- 28) Methyl cyclopentane 430
- 29) Dichloropropane 3800
- 30) Hexane 260
- 31) Methyl cyclohexane 420
- 32) Trichloropropane 370
- 33) Ethyl benzene 1900
- 34) Thiophene 200
- 35) 2-butanone 250
- 36) 2-butanone 440
- 37) Bromodichloromethane 1300
- 38) Bromochloropropene 1200
- 39) Bromochloropropane 320
- 40) Trichloroethanol 670
- 41) Methyl cyclohexane 200
- 42) Cyclopentene 380
- 43) Methyl cyclohexane 220
- 44) Cyclopentene 270
- 45) Methyl cyclohexane 490
- 46) Hexane 240
- 47) Methyl cyclohexane 390
- 48) Thiophene 320
- 49) Dichloropropane 380
- 50) Ethyl benzene 200
- 51) Ethyl benzene 440

All concentrations are in $\mu\text{g/l}$ (ppb)

- 52) Dipropyl ether 250
- 53) Thiophene 240
- 54) Ethyl benzene 210
- 55) dipropyl ether 360
- 56) Methyl cyclopentane 590
- 57) Methyl cyclohexane 540
- 58) Ethyl benzene 730
- 59) Bromodichloromethane 87001)
- 60) Dibromochloromethane 450
- 61) Ethyl benzene 400
- 62)



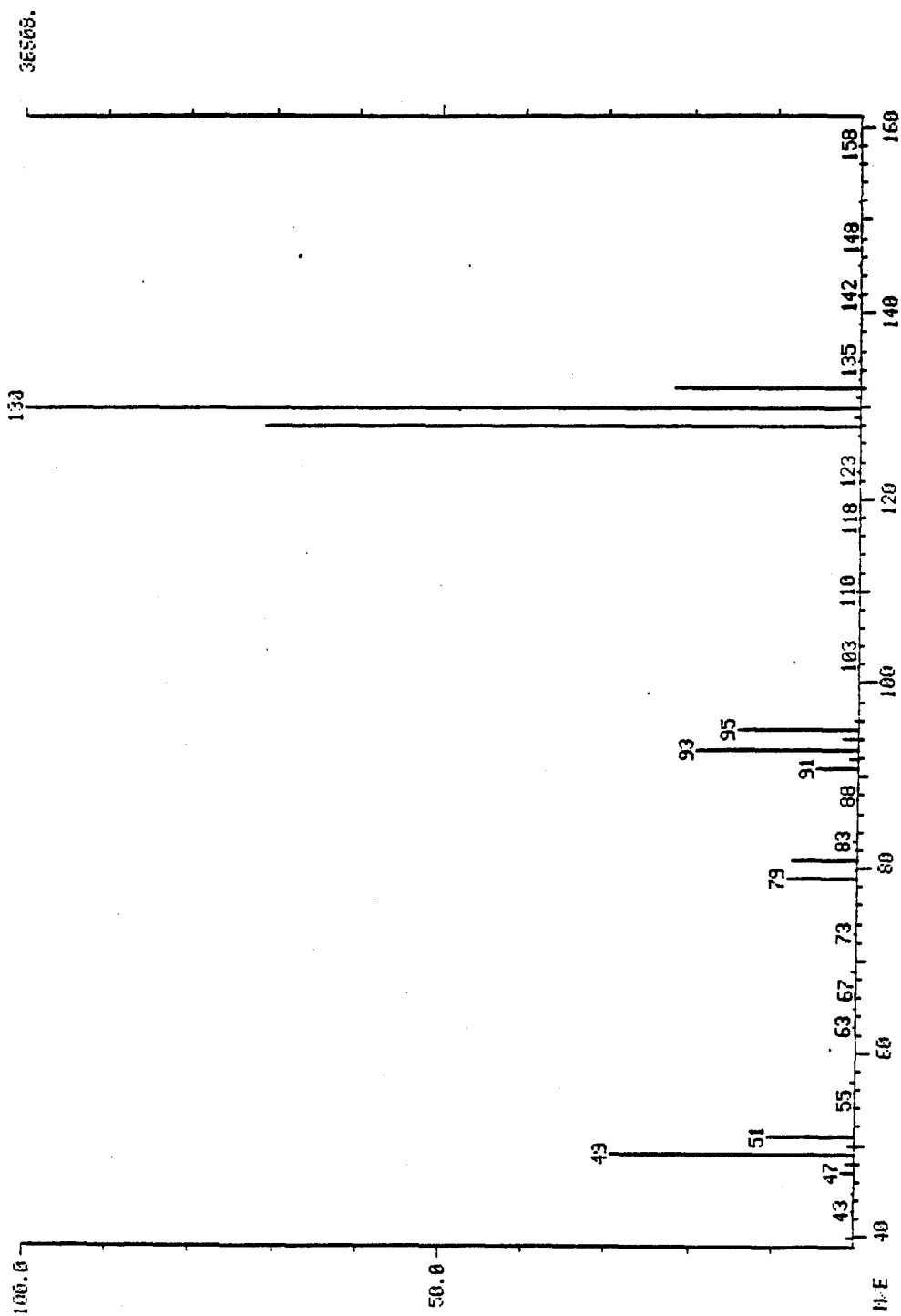
- 63) Dibenzyl ether 350
- 64) From packed column BN run
- 65) From capillary column BA run
- 66) Ethyl benzene 240
- 67) Ethyl benzene 1000
- 68) Dichloroethylene 970
- 69) Methyl cyclohexane 580
- 70) Carbon disulfide 360
- 71) Cyclopentene 230
- 72) Cyclohexane 210
- 73) Thiophene 550
- 74) Methyl cyclohexane 610
- 75) Ethyl benzene 310
- 76) Diphenyl ether 380
- 77) VOA sample taken from quart jar
- 78)

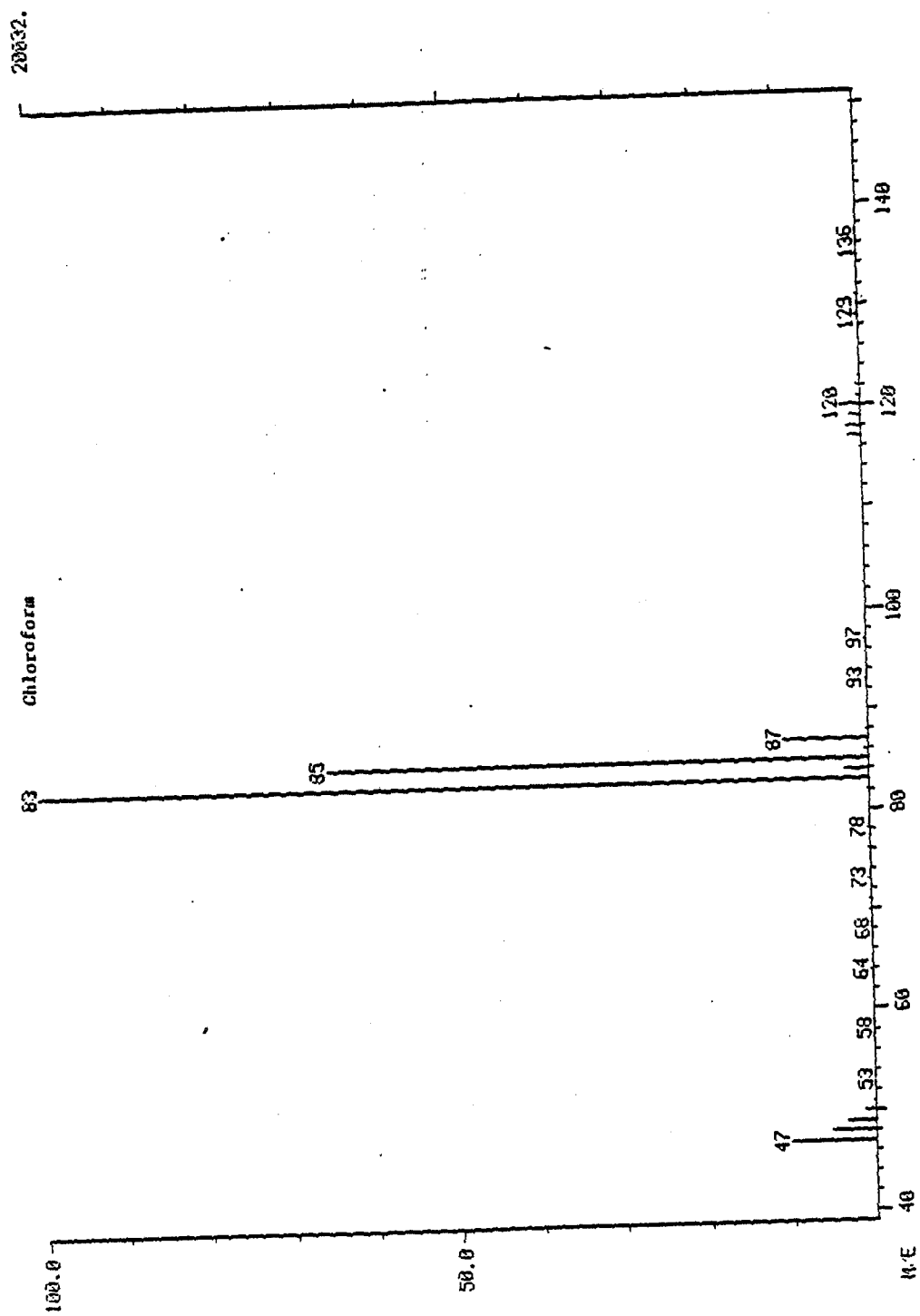


- 79) Cyclopentene 360
- 80) Cyclohexane 260
- 81) Hexane 400
- 82) Methyl cyclohexane 450
- 83) Ethyl benzene 490
- 84) Dipropyl ether 320
- 85) Sample lost
- 86) RIR-24-642-79
- 87) RIR-24-643-79
- 88) Average of two determinations
- 89) Trichlorophenol 910

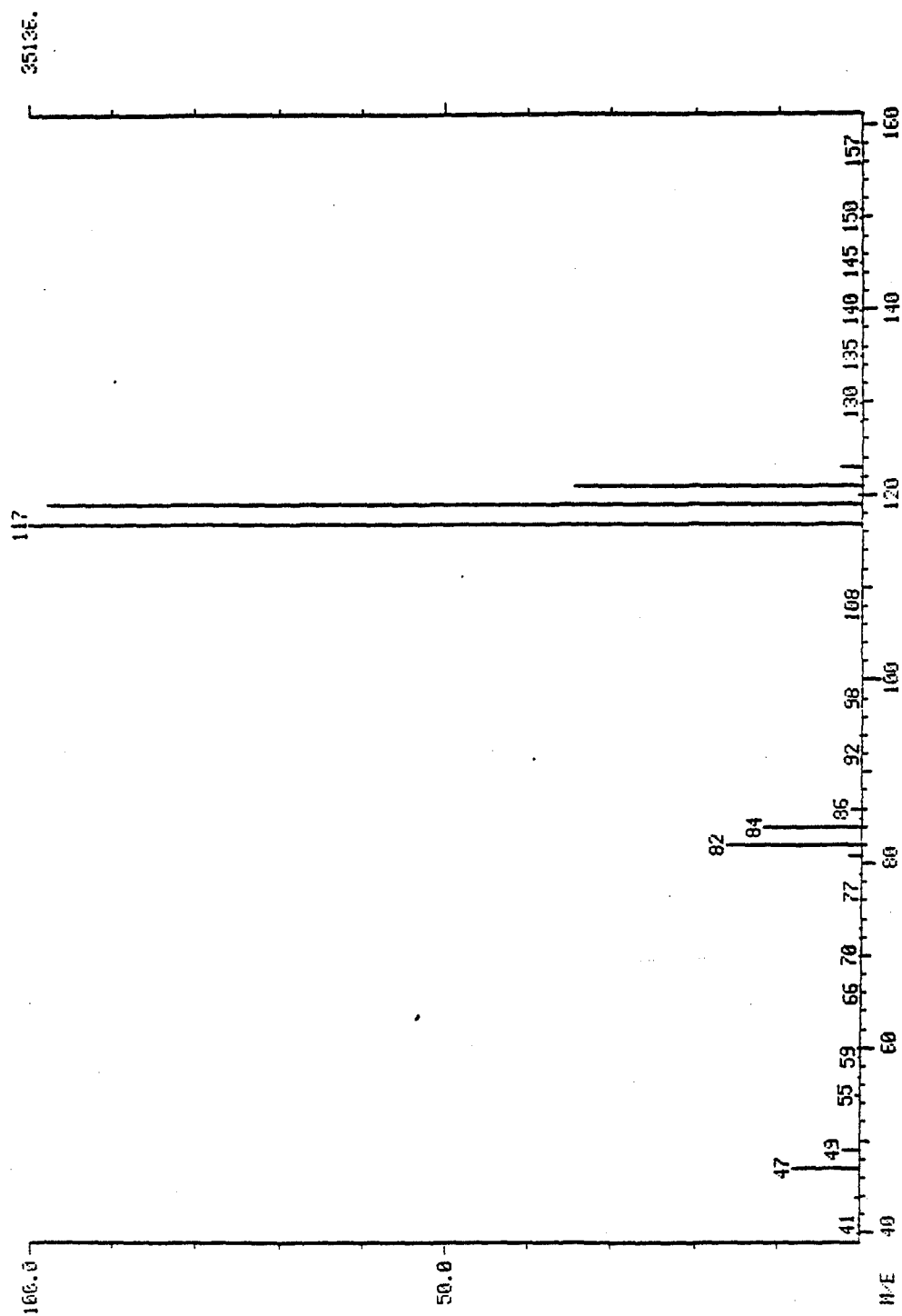
All concentrations are in $\mu\text{g/l}$ (ppb)

Bromochloromethane
Internal Standard

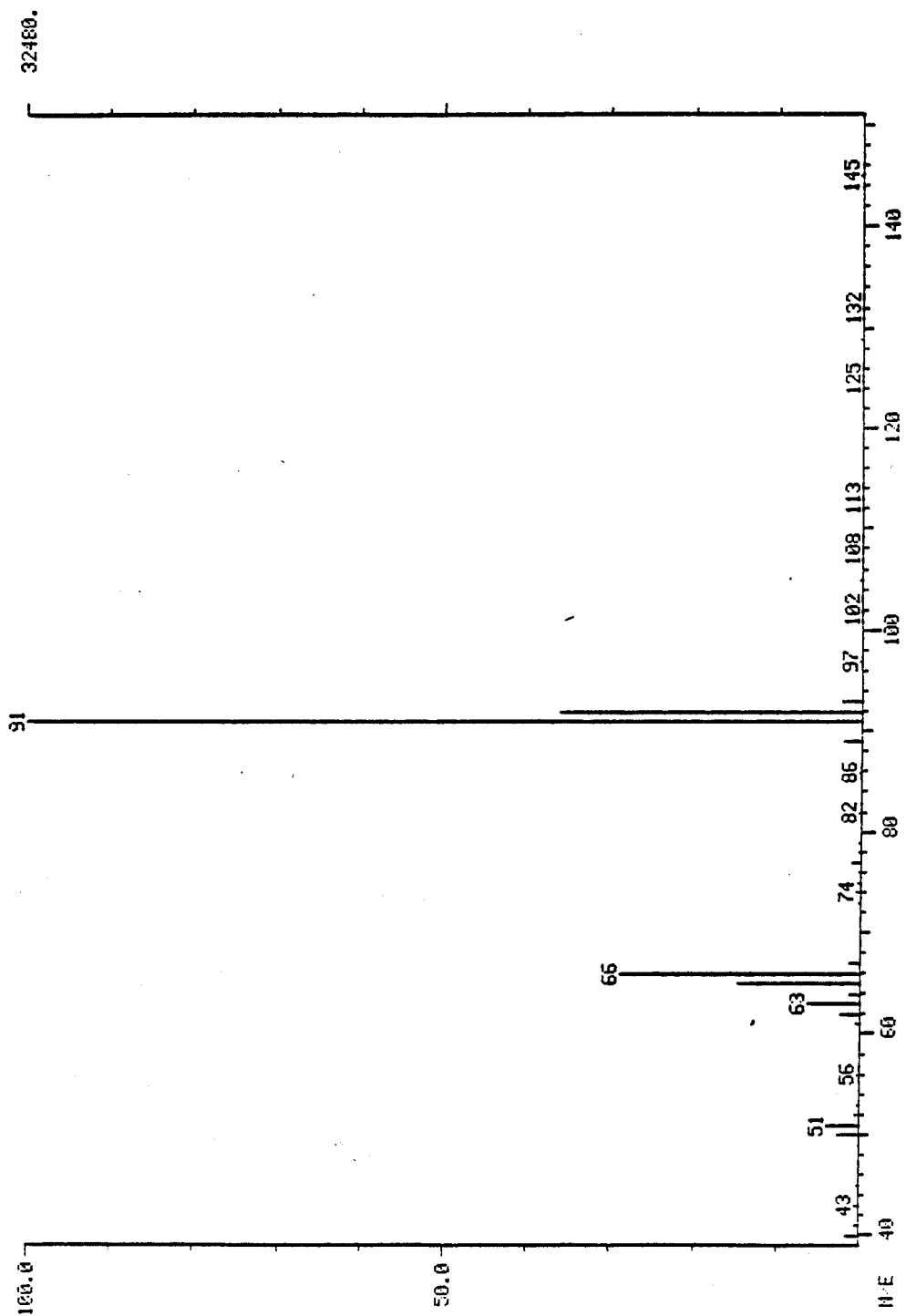


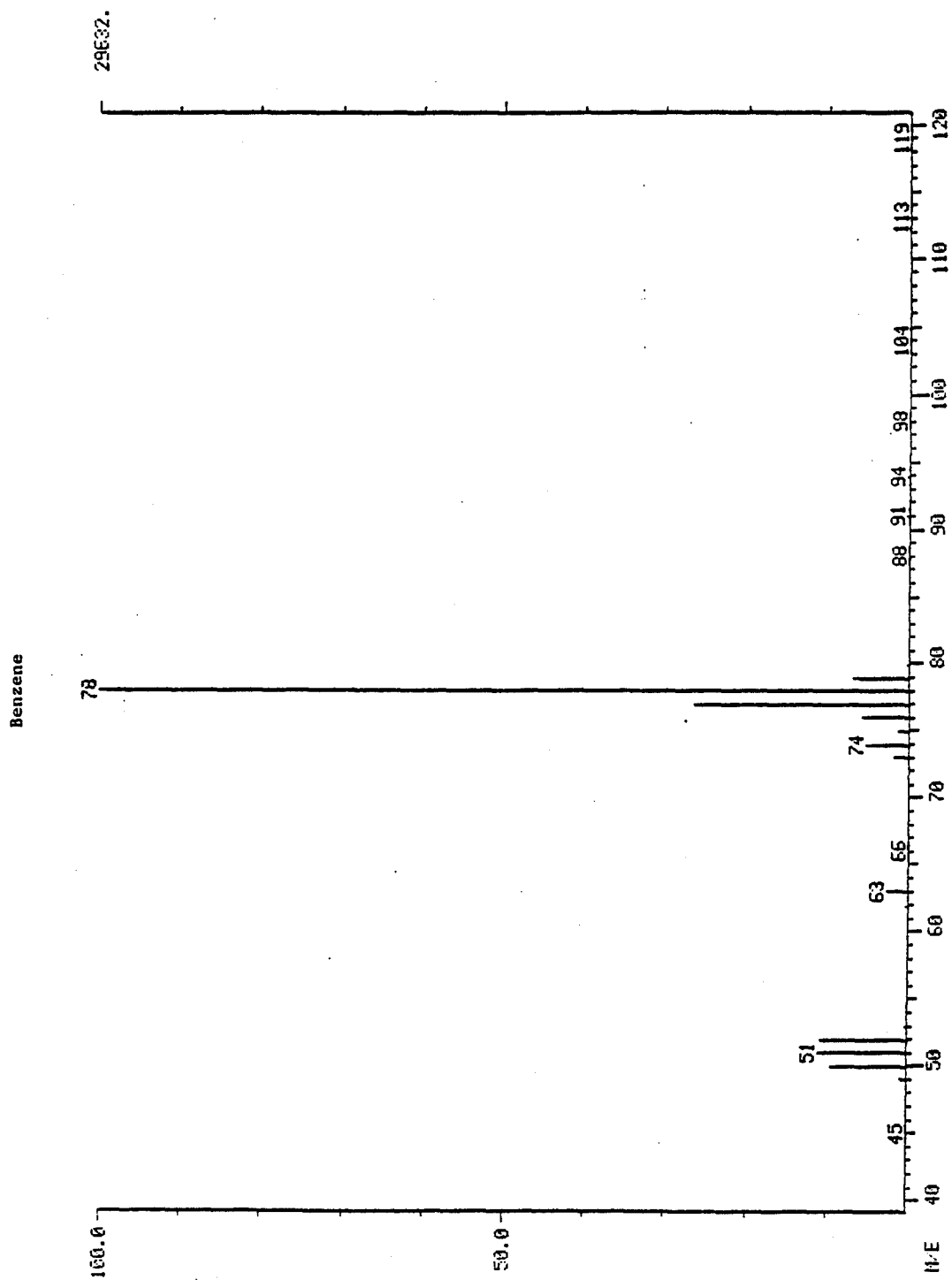


Carbon Tetrachloride

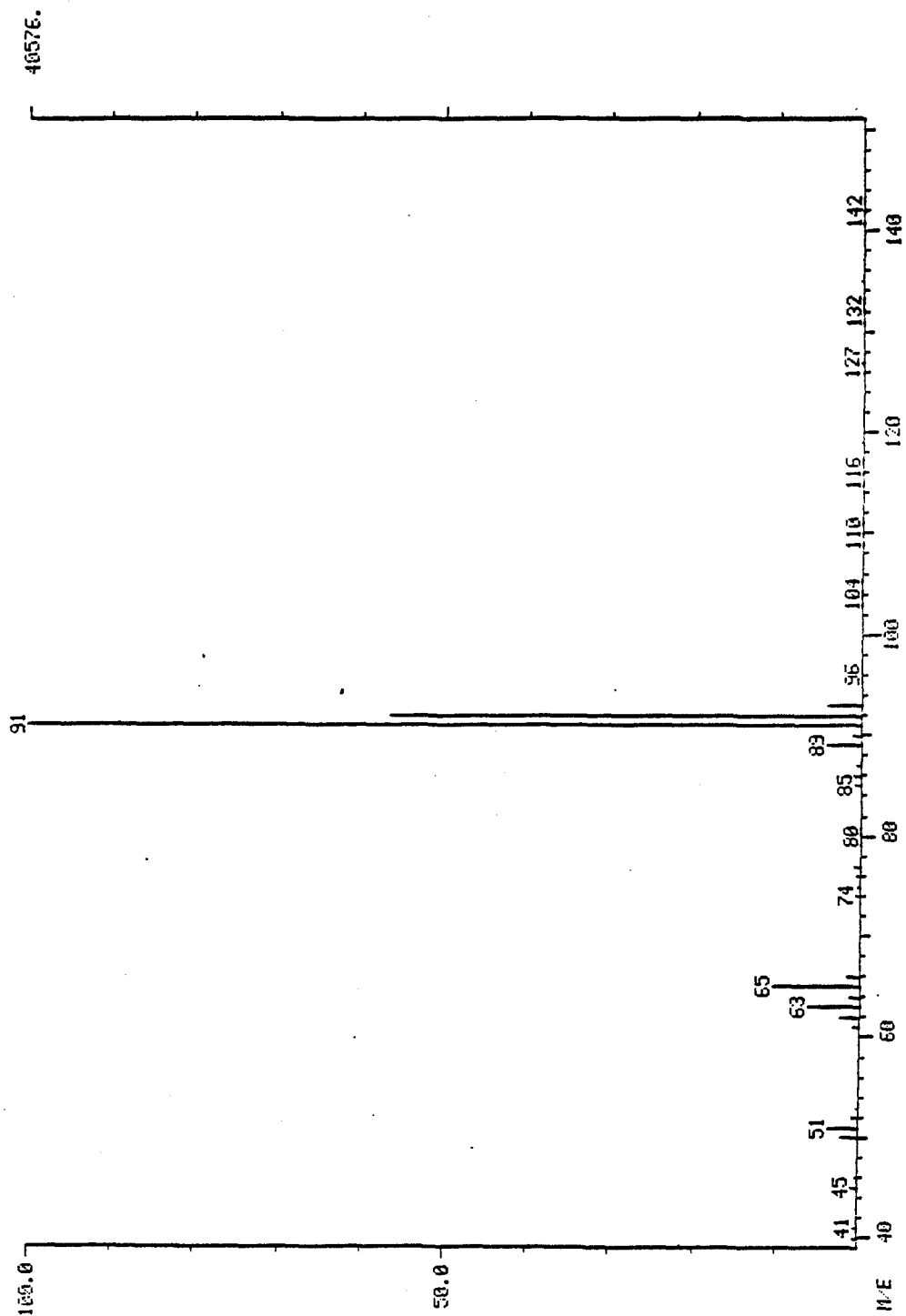


Bicycloheptadiene

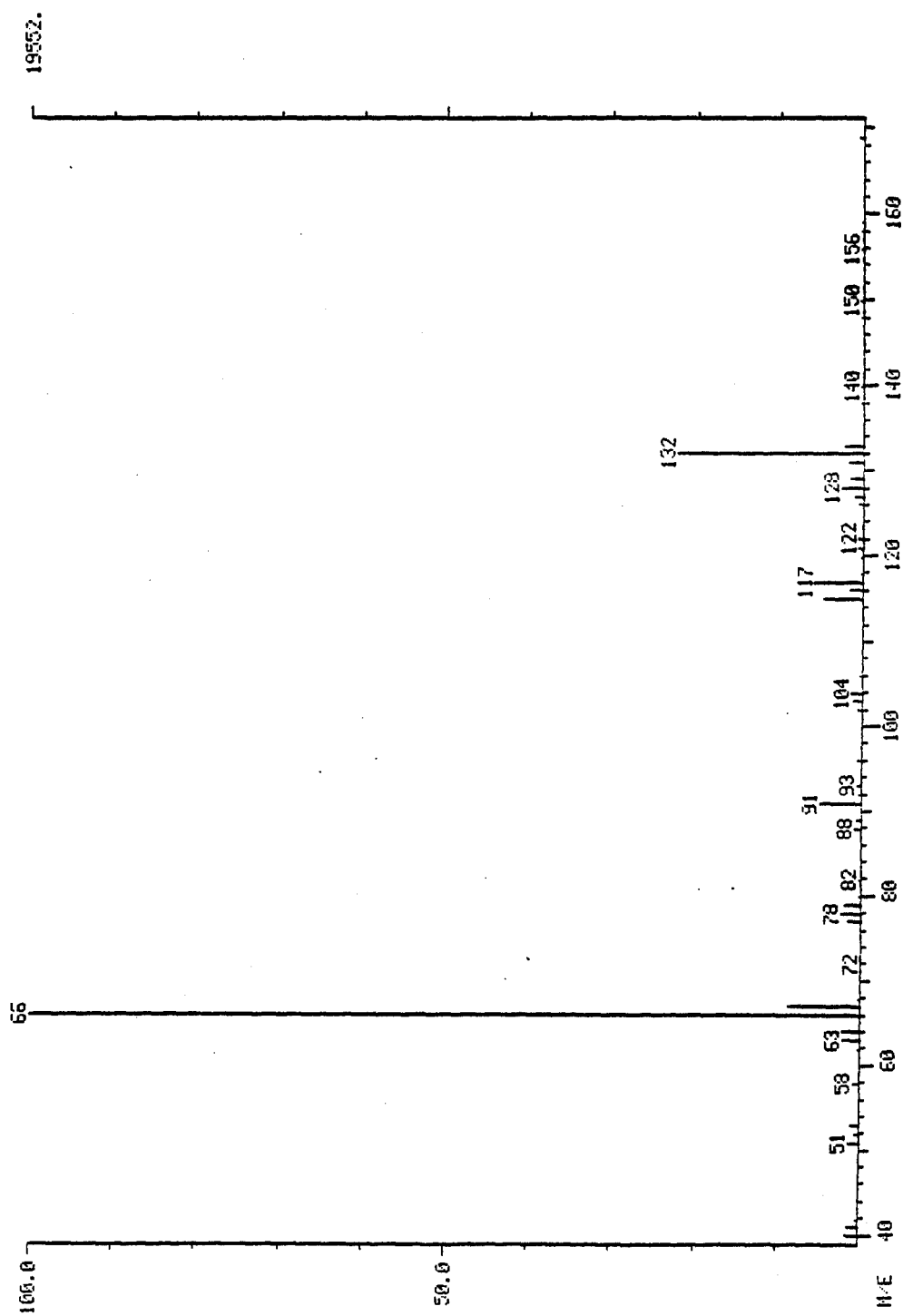


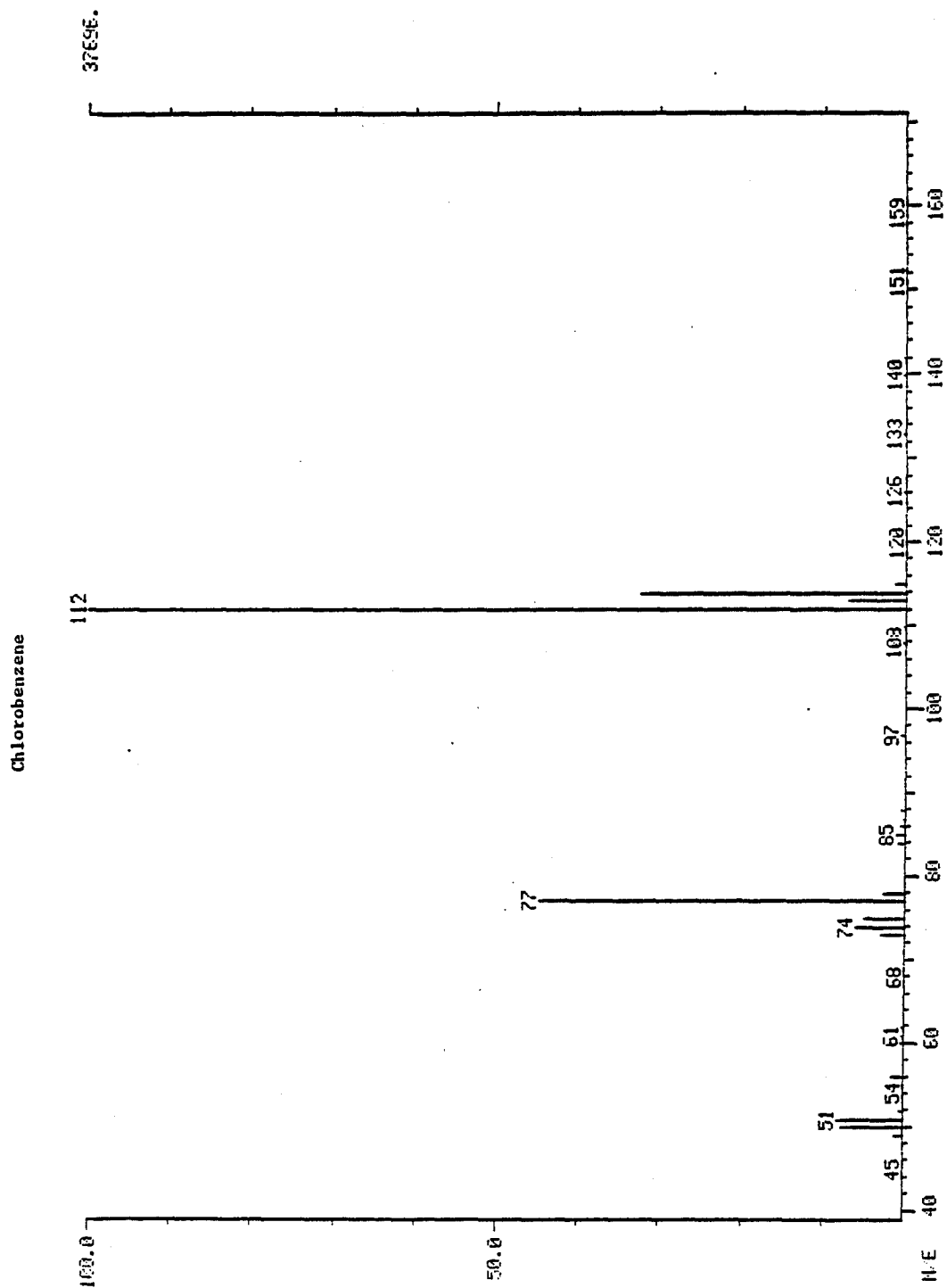


Toluene

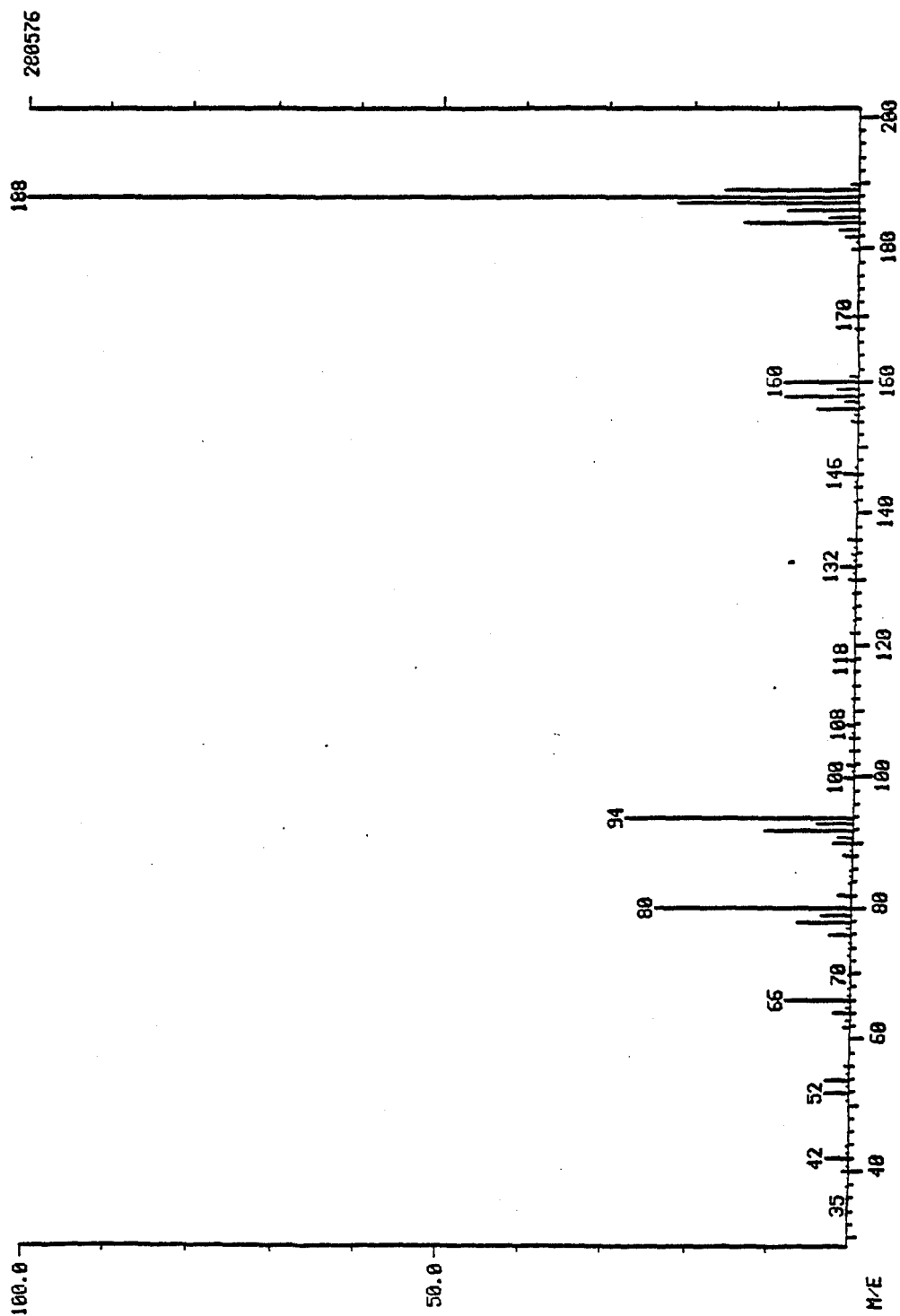


Dicyclopentadiene

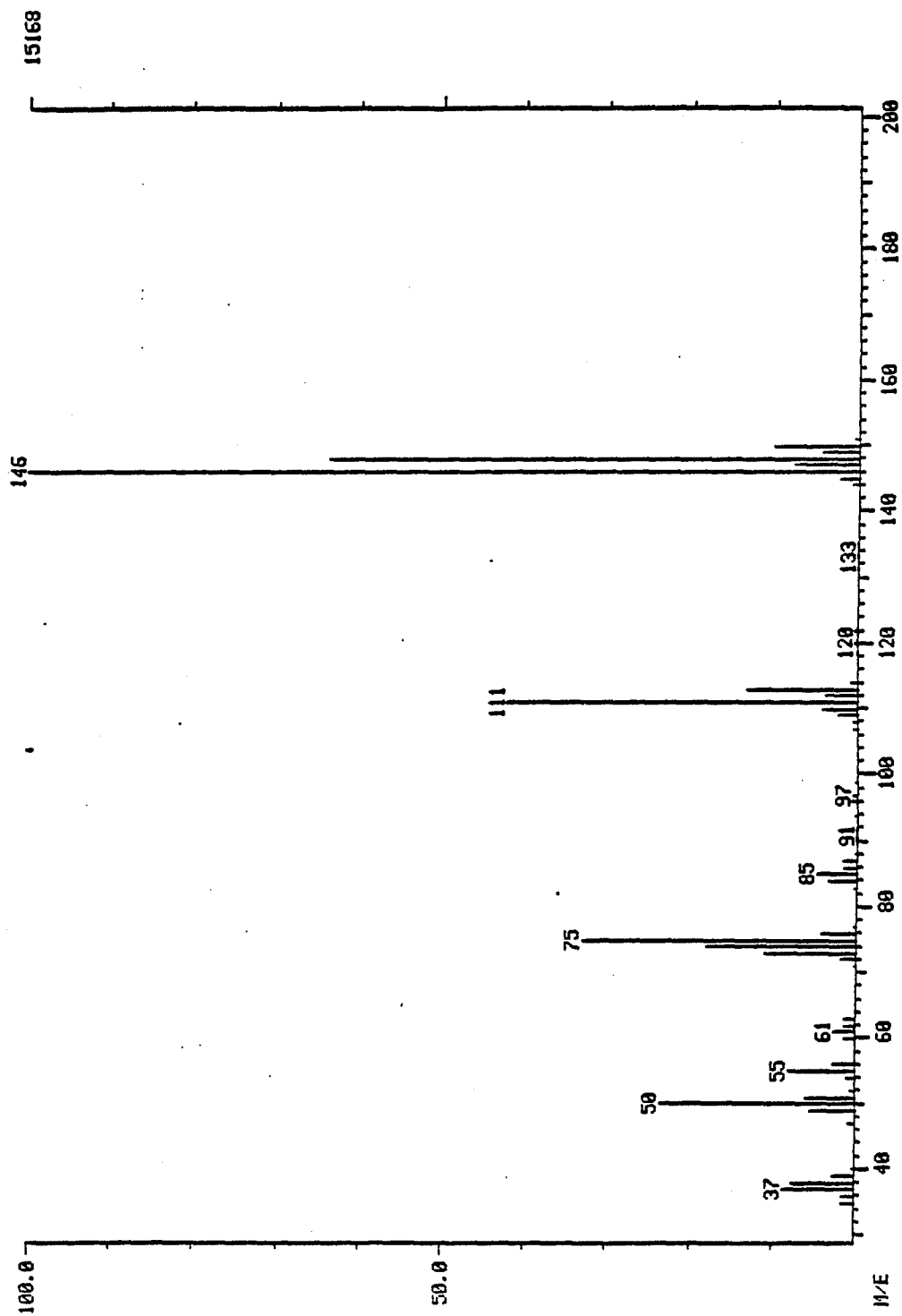




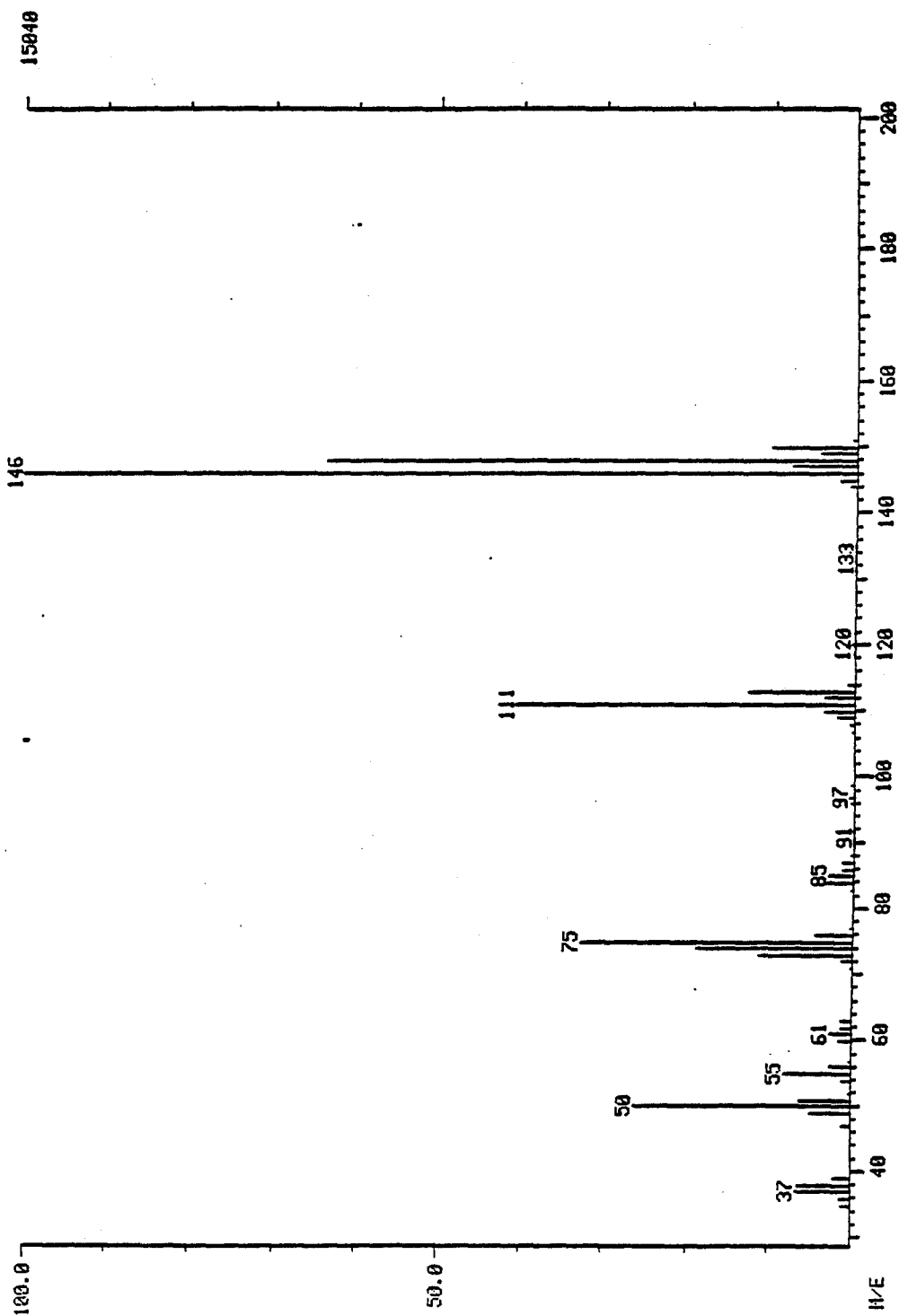
Dio-Anthracene



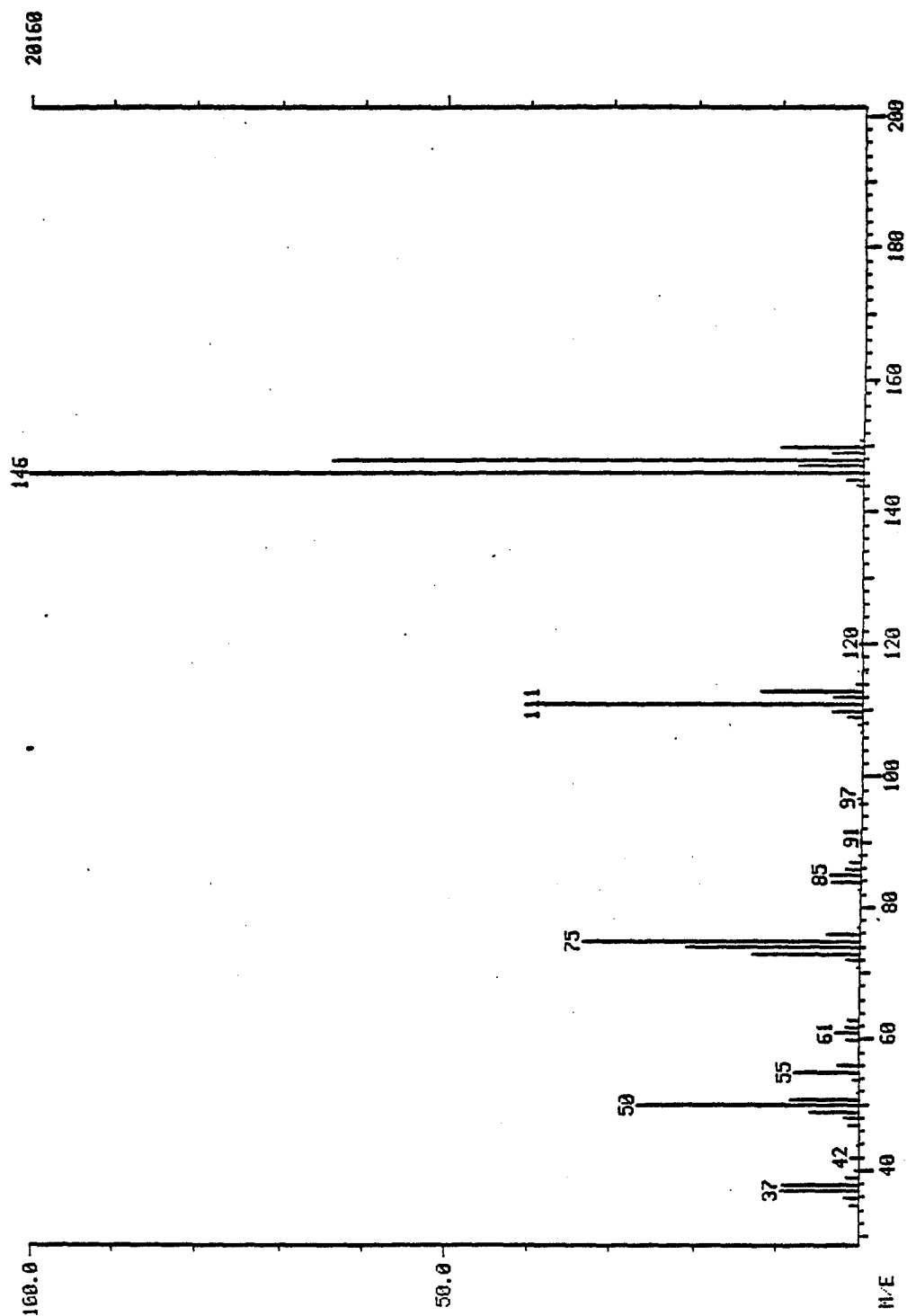
O-DCB



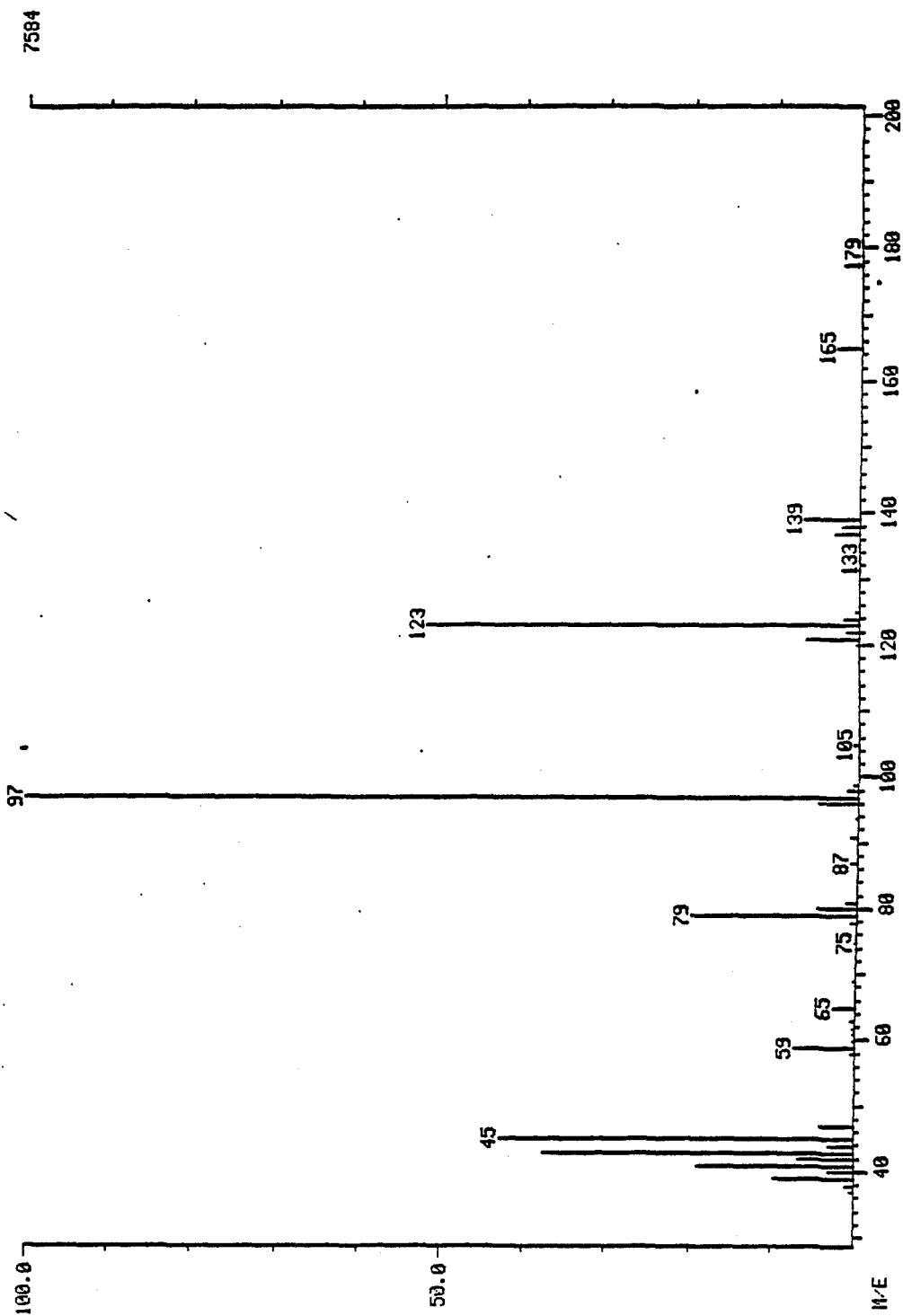
M-DCB



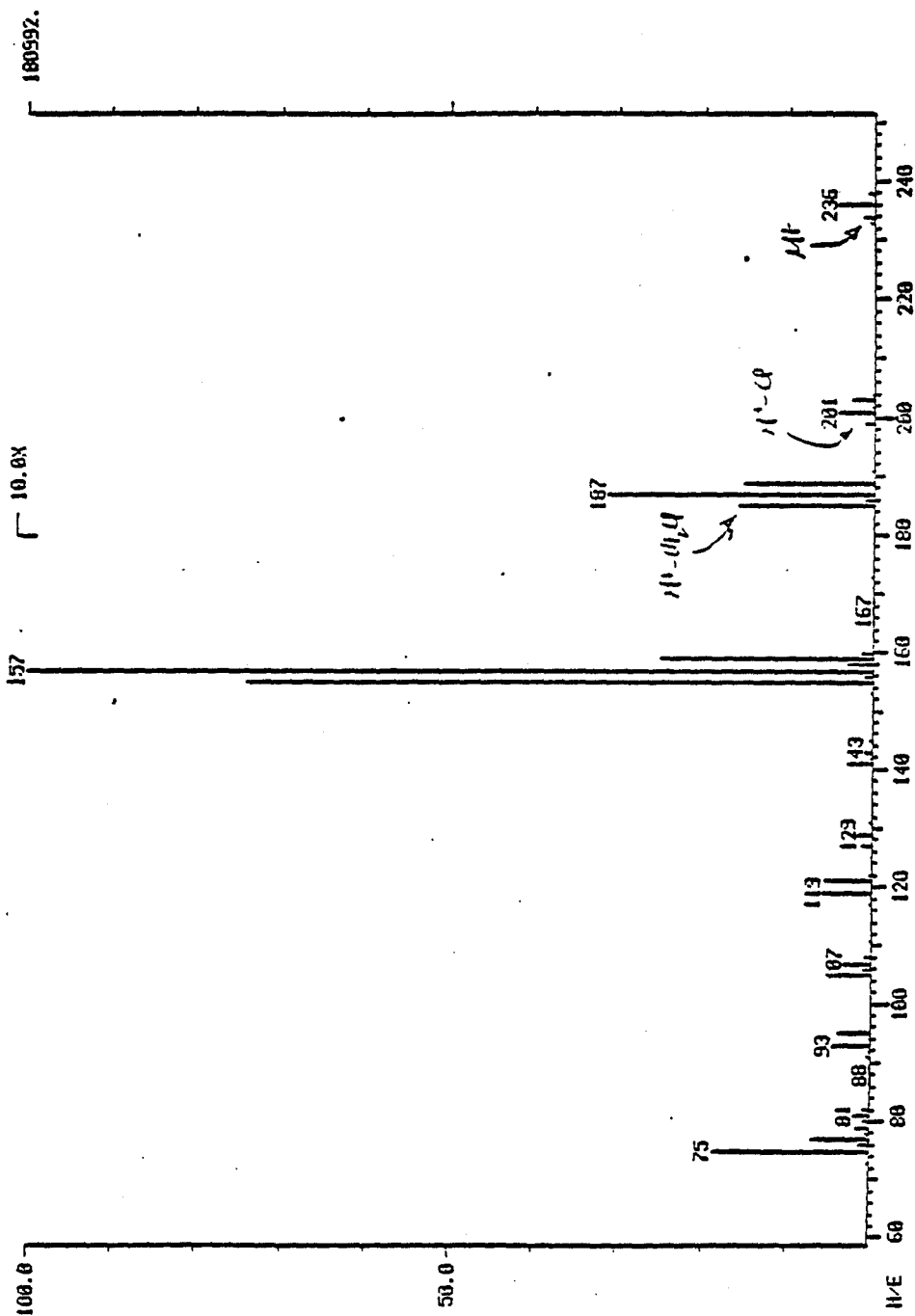
P-DCB



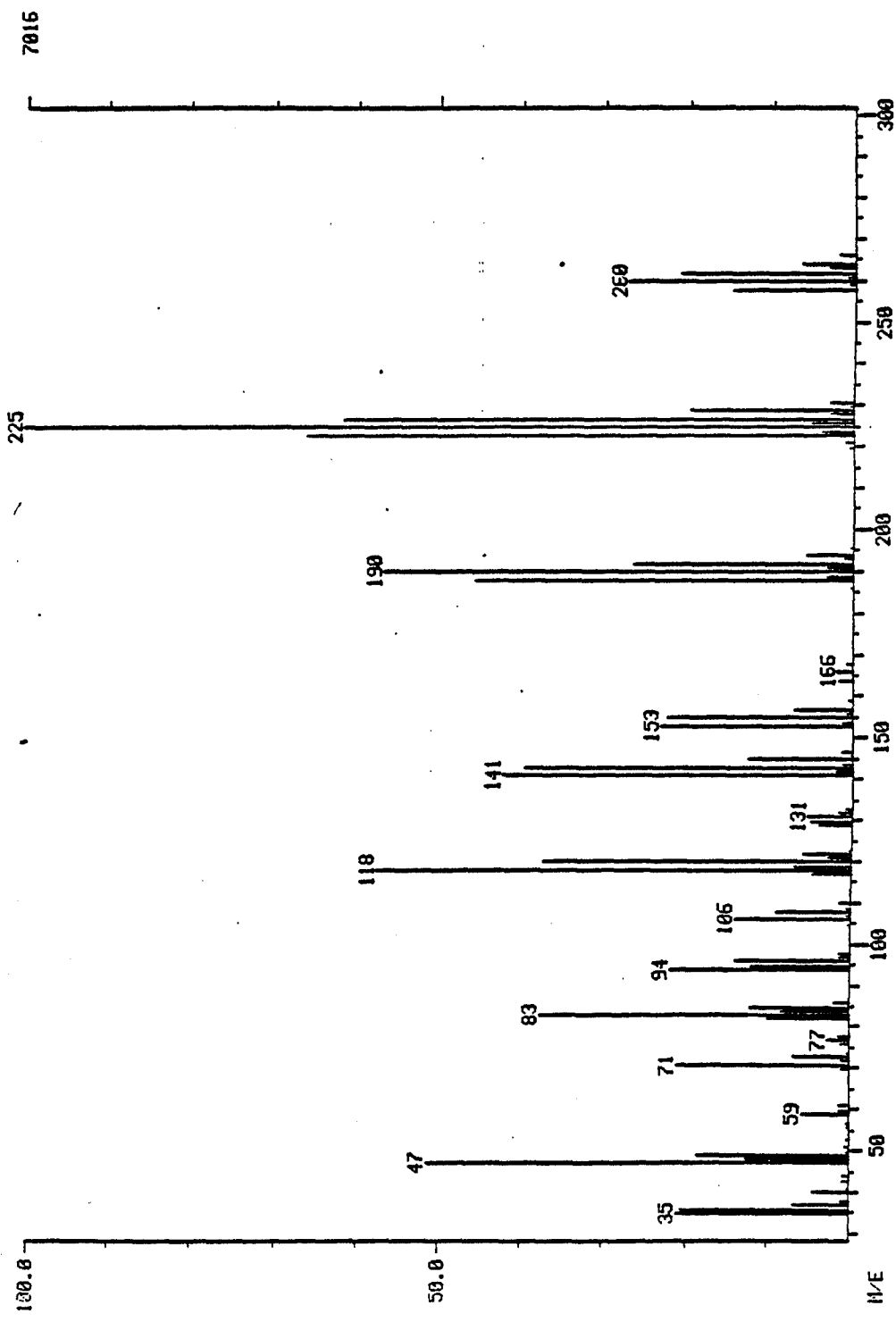
DIMP



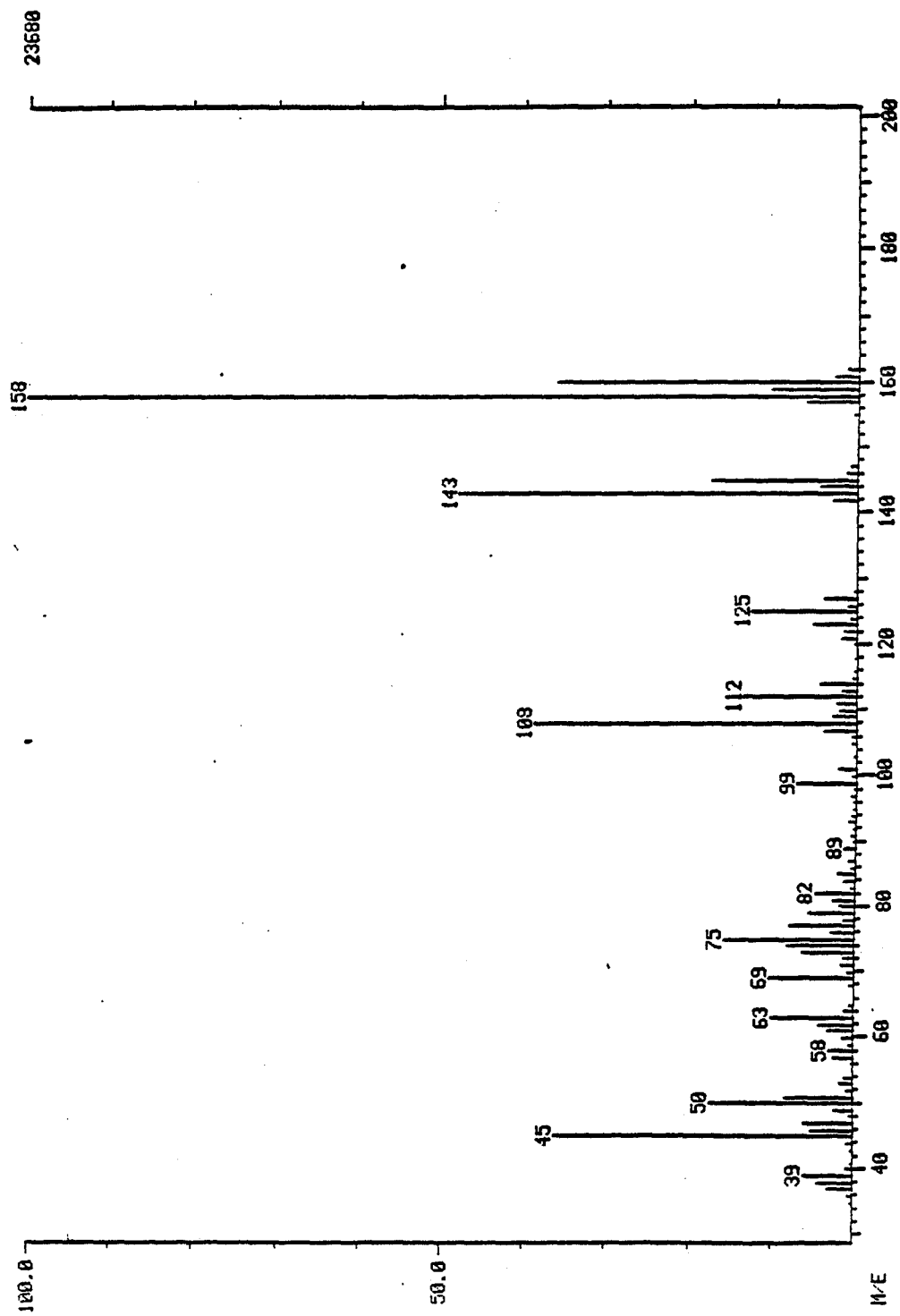
DBCP



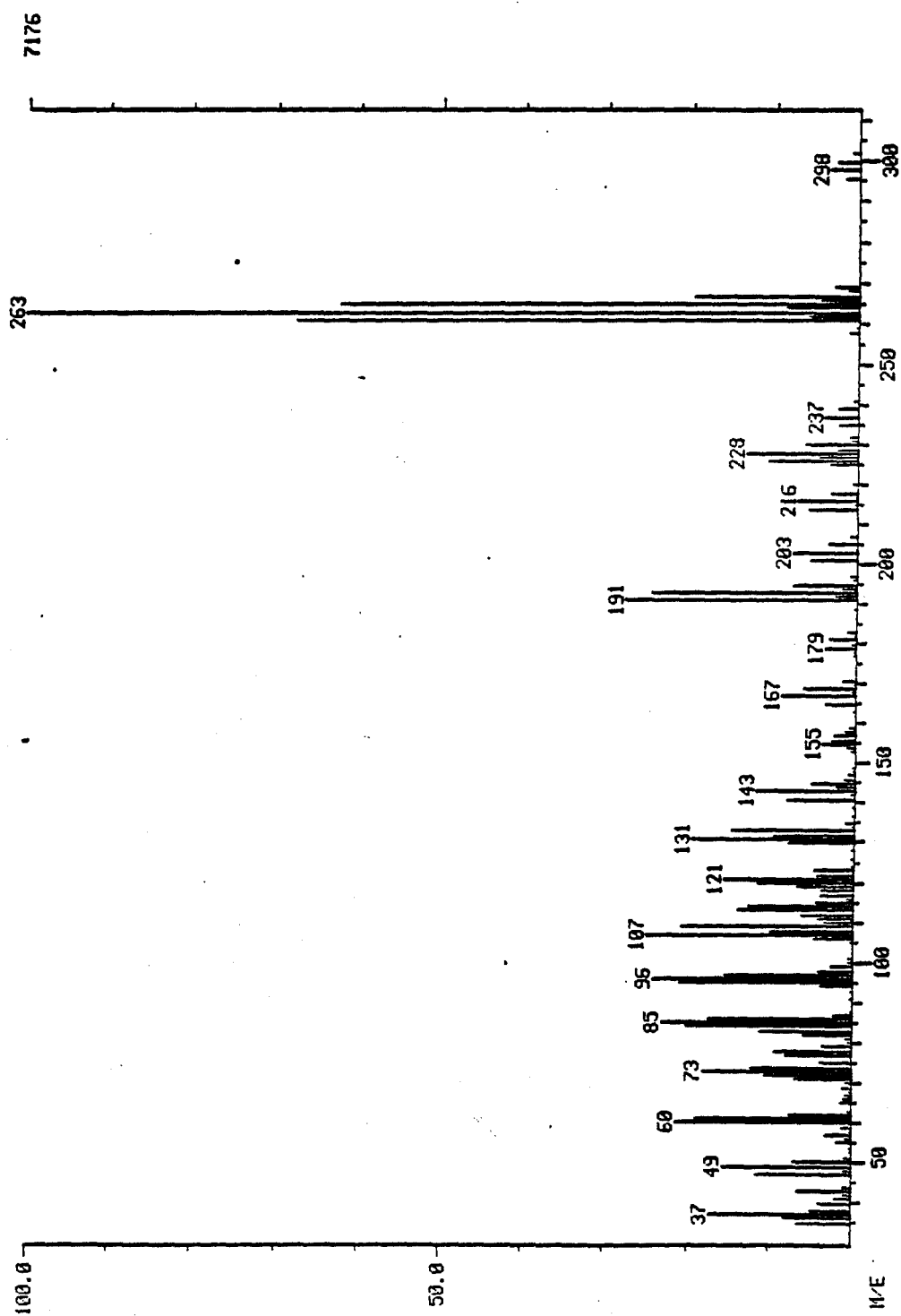
HCBD



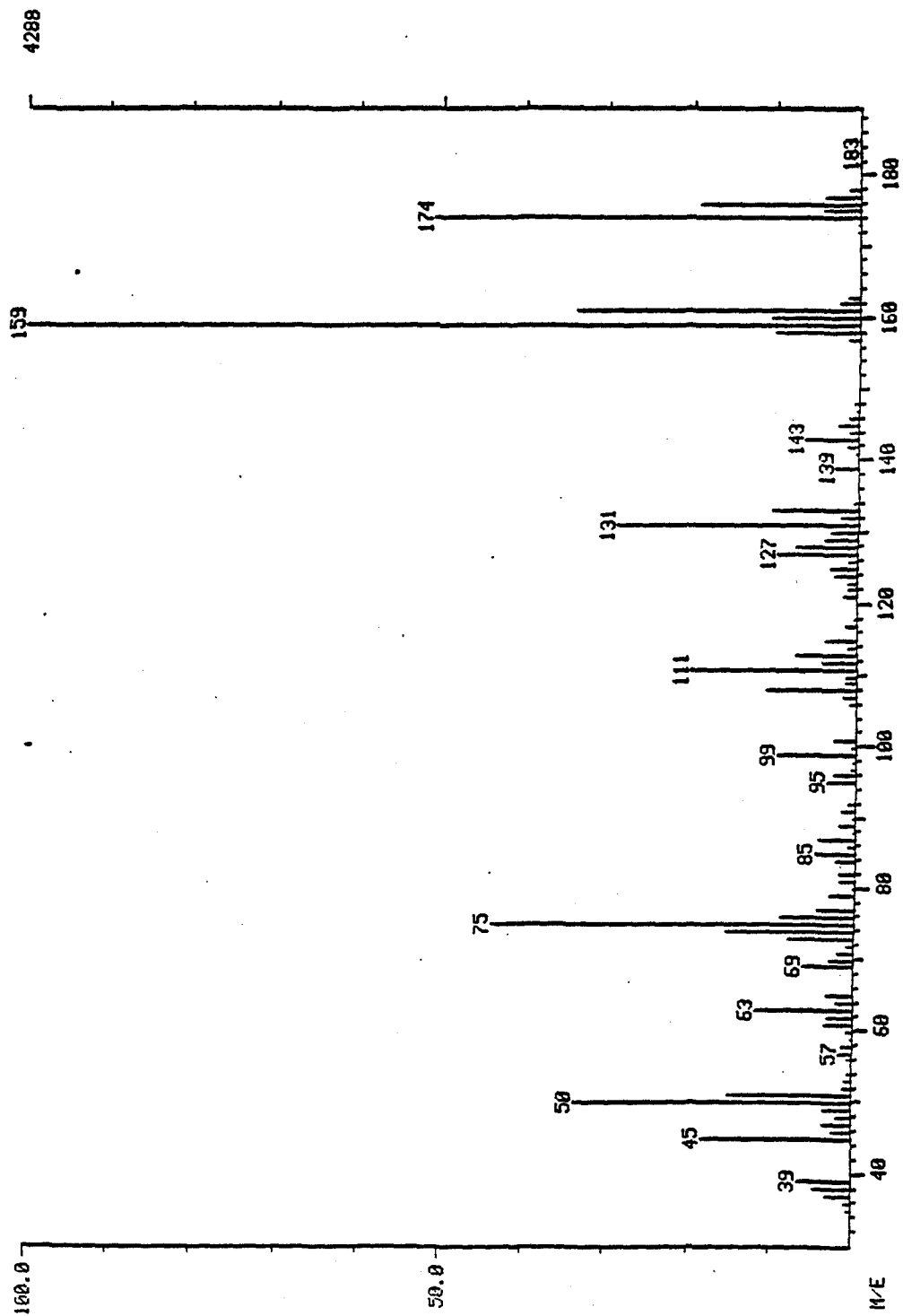
SD 9636



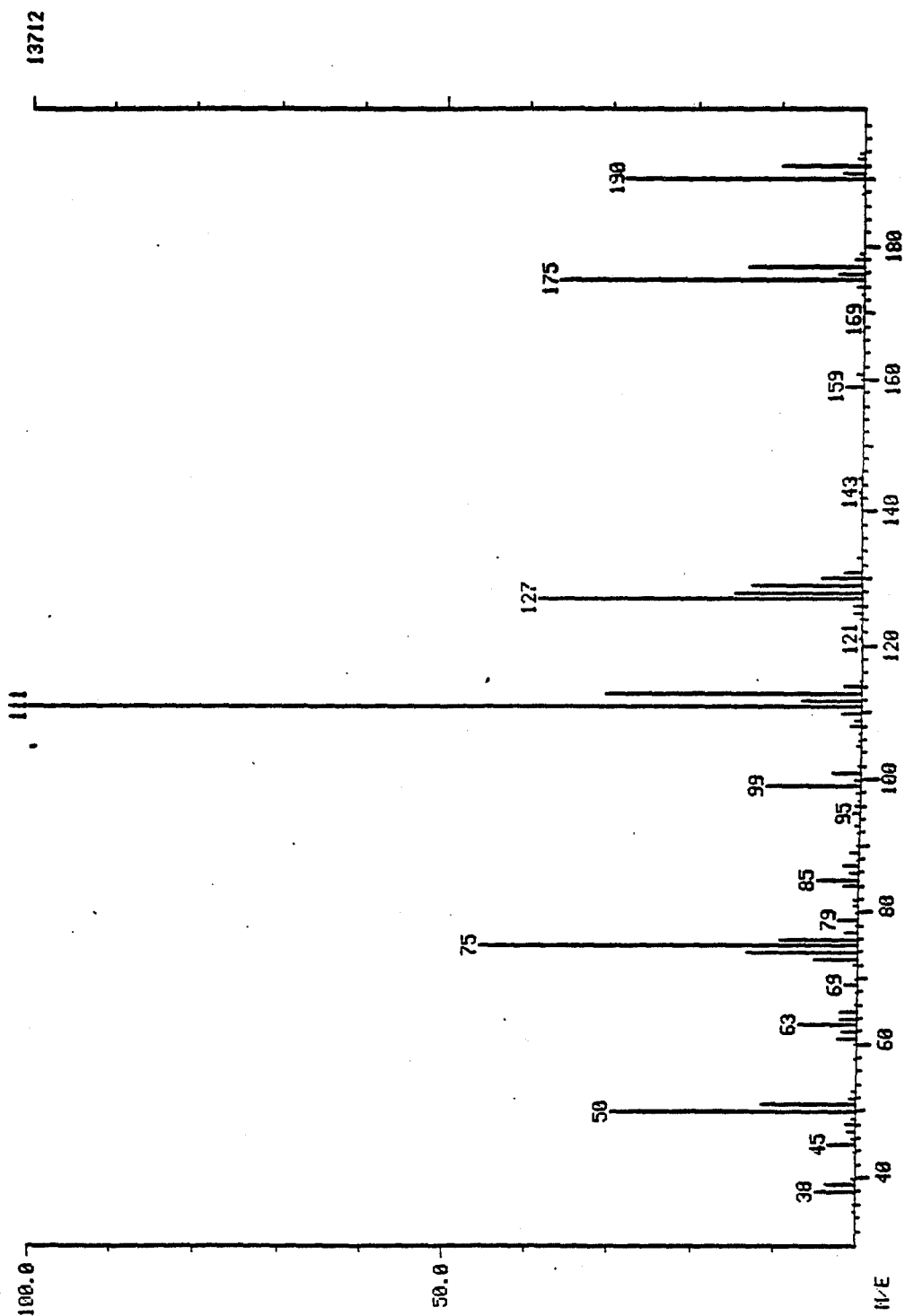
HCNB



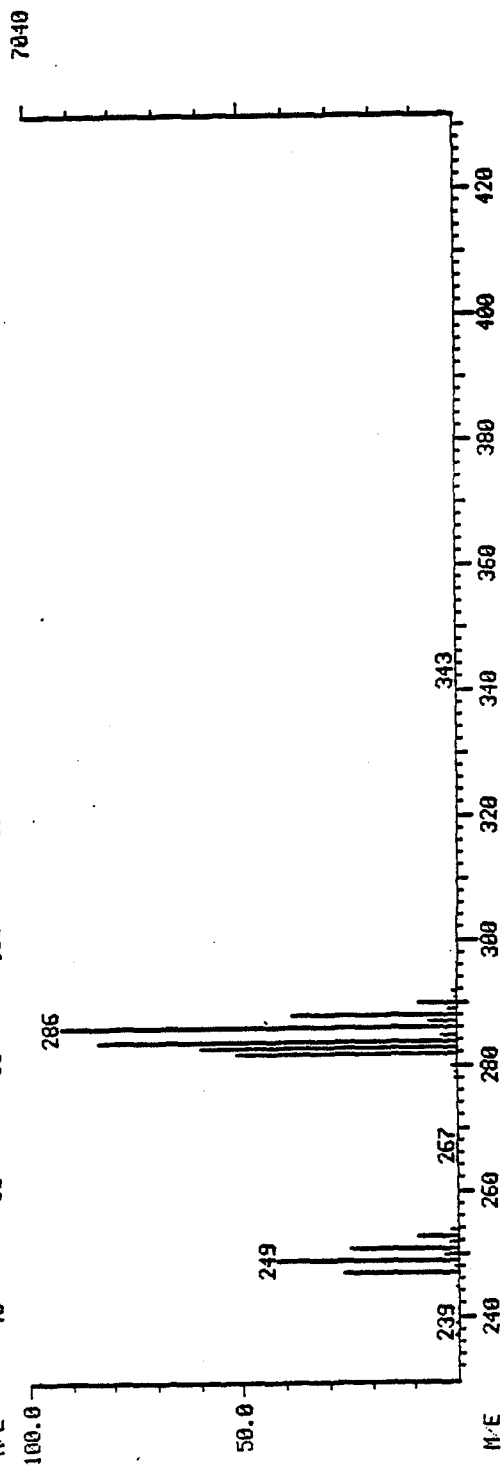
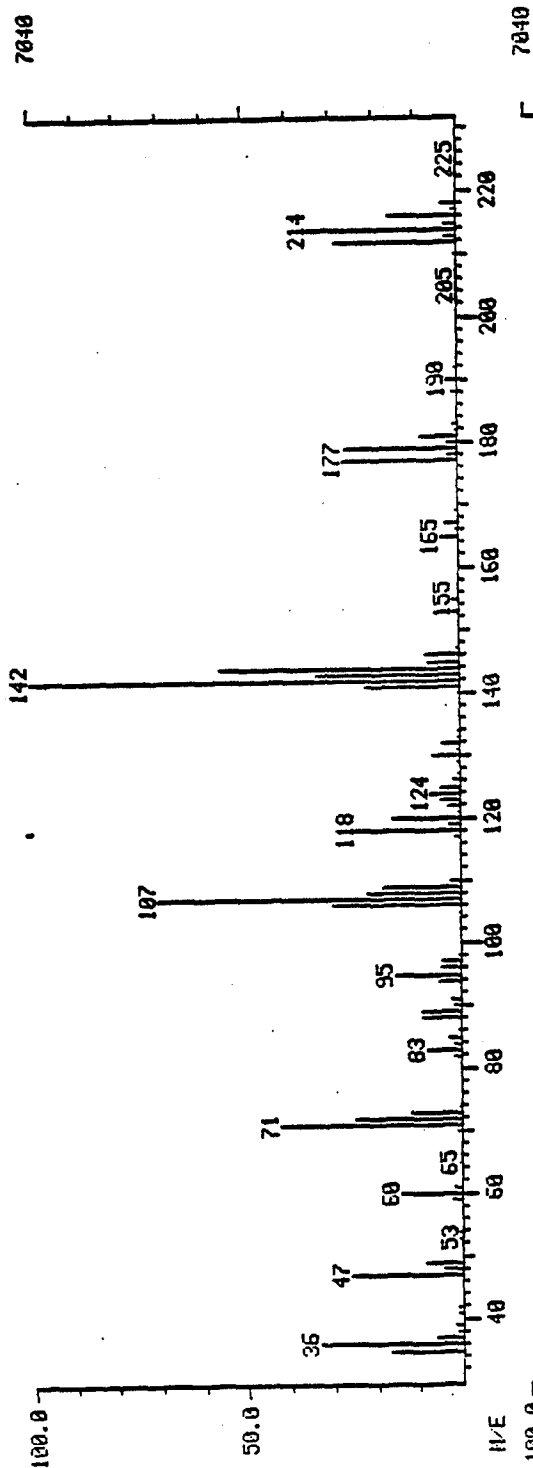
SD 13957



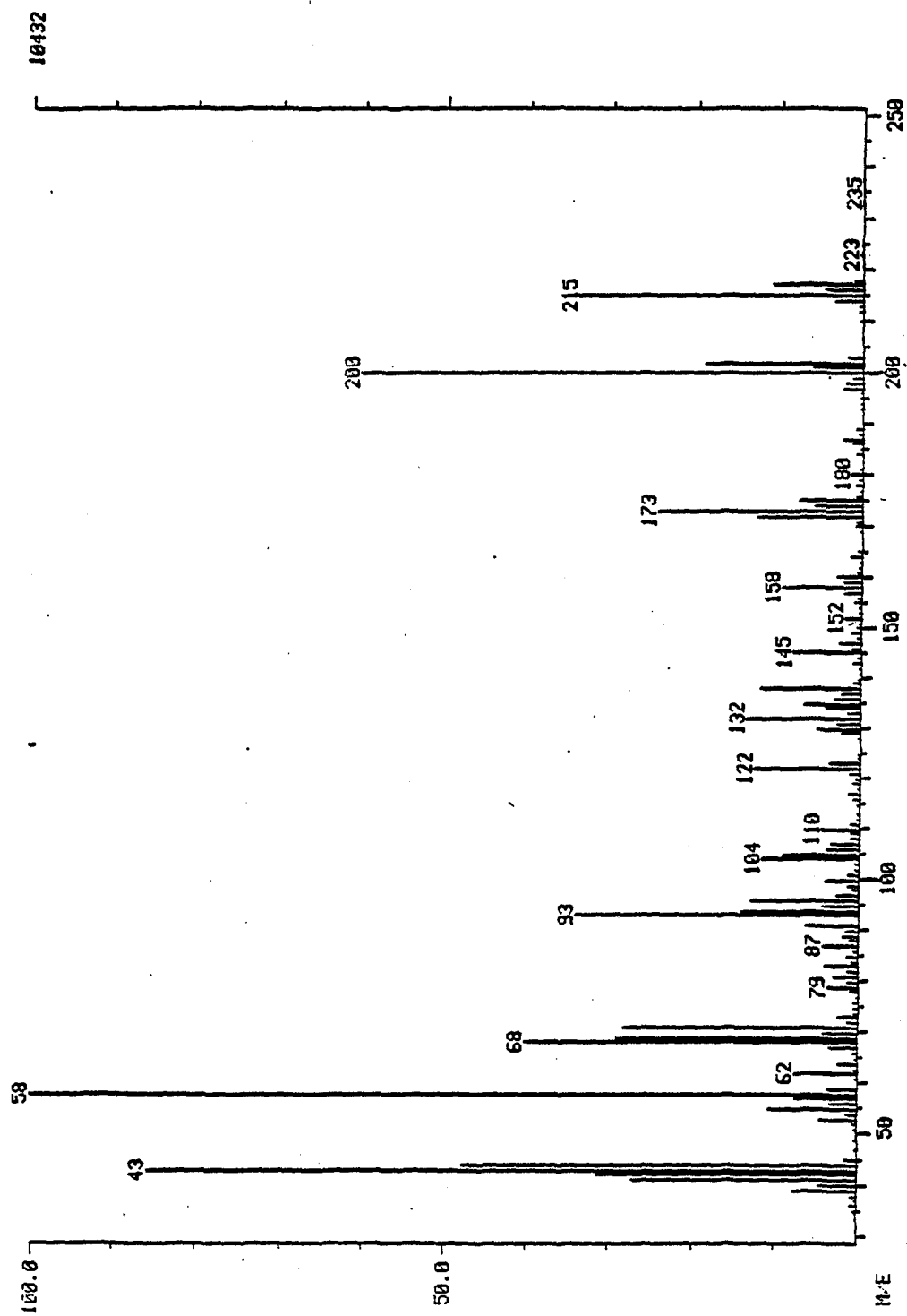
SD 13000



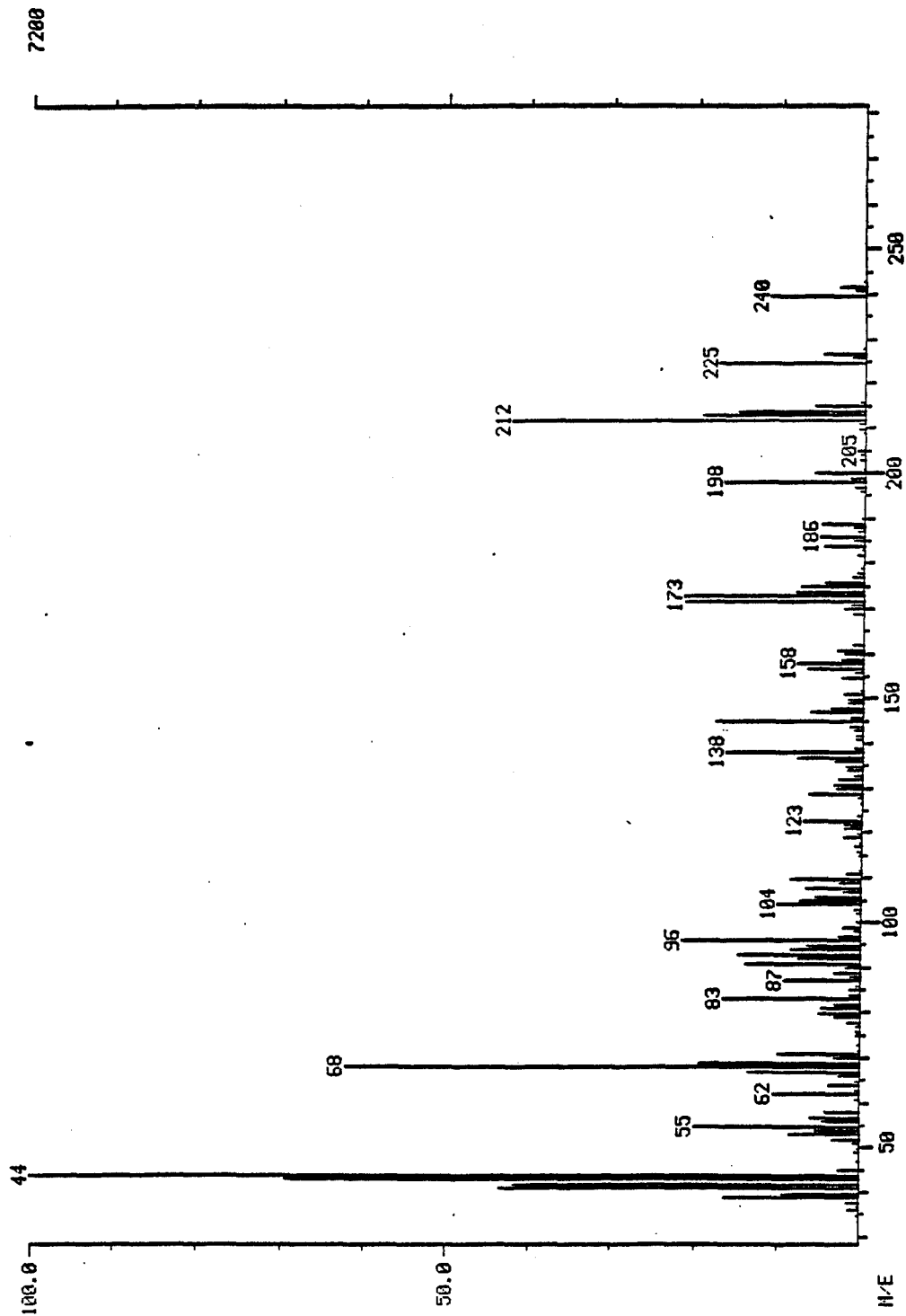
HCB



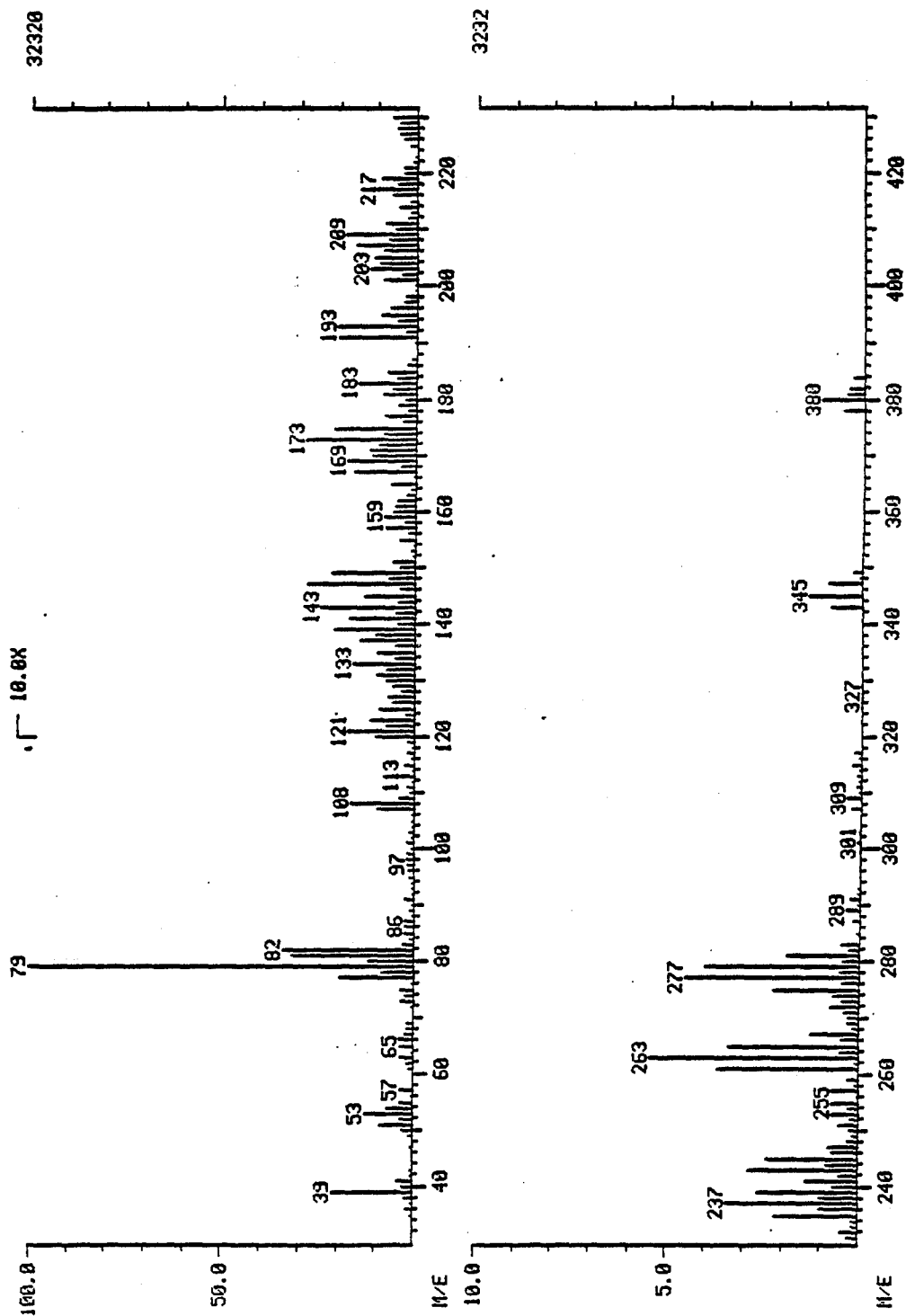
Atrazine



BLADEX



HEAD



2800' 1500' W 2000' W 1800' W 1000' W 500'

SECTION 35

013
○

PARKING L

SECTION 2

94

75.15

BOLT STORAGE

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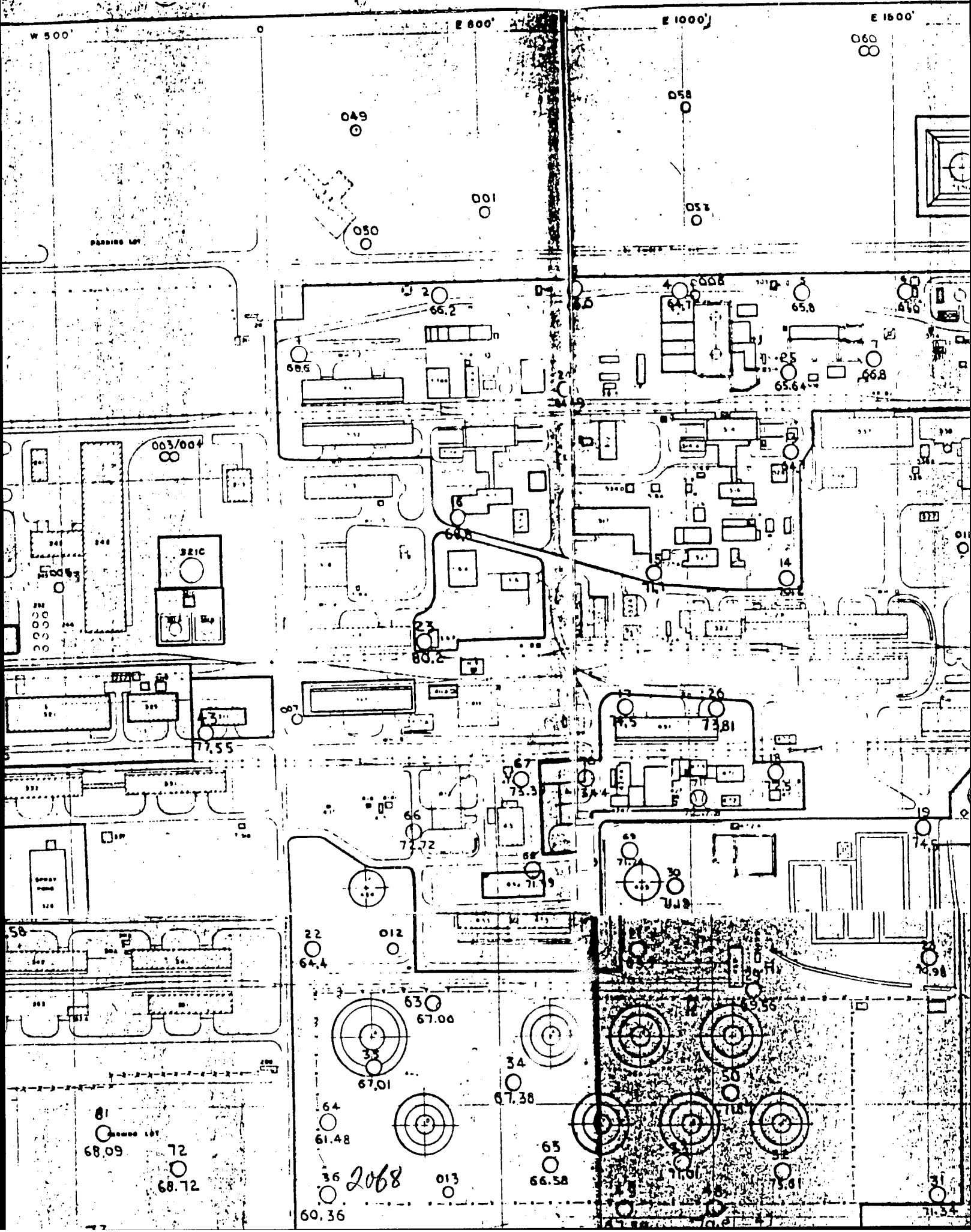
006

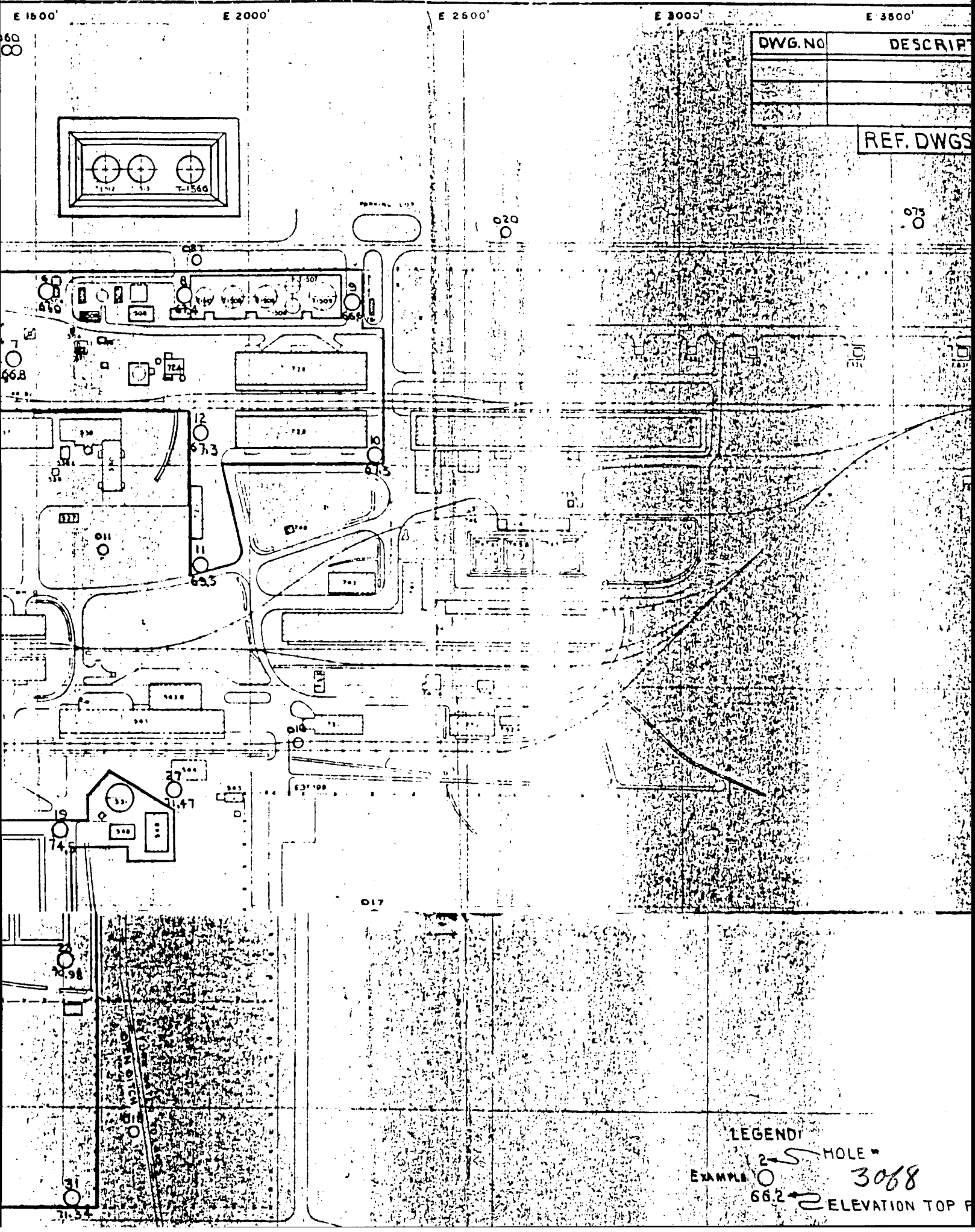
006

1068

81
○
68.09

73








DWG. NO.	DESCRIP.

REF. DWGS.

LEGEND

EXAMPLE  HOLE
 3068
 66.2  ELEVATION TOP

E 2500'

E 3000'

E 3500'

N 800'

DWG. NO	DESCRIPTION

REF. DWGS.

SECTION 35

SECTION 1

S 1000'

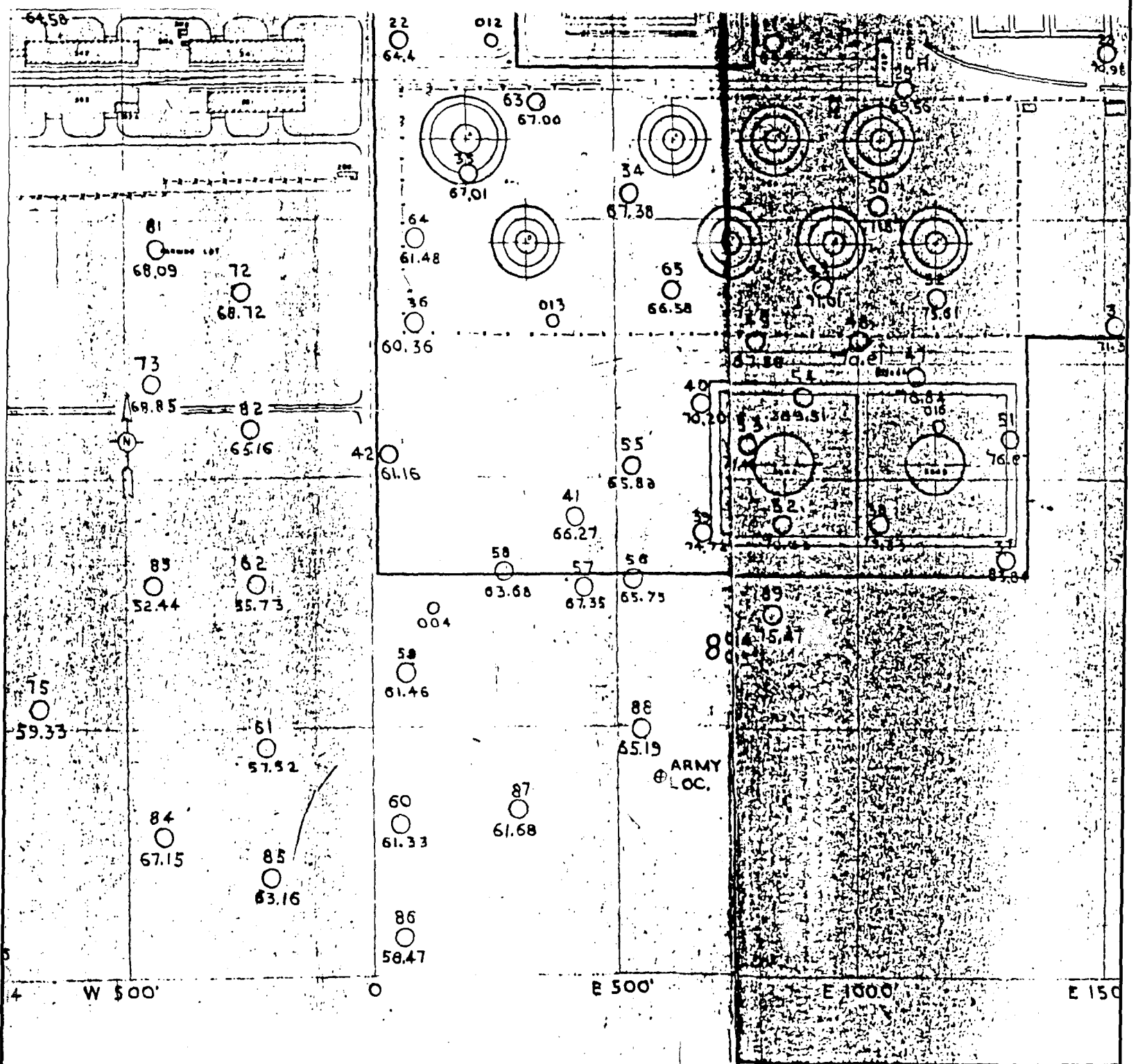
S 2000'

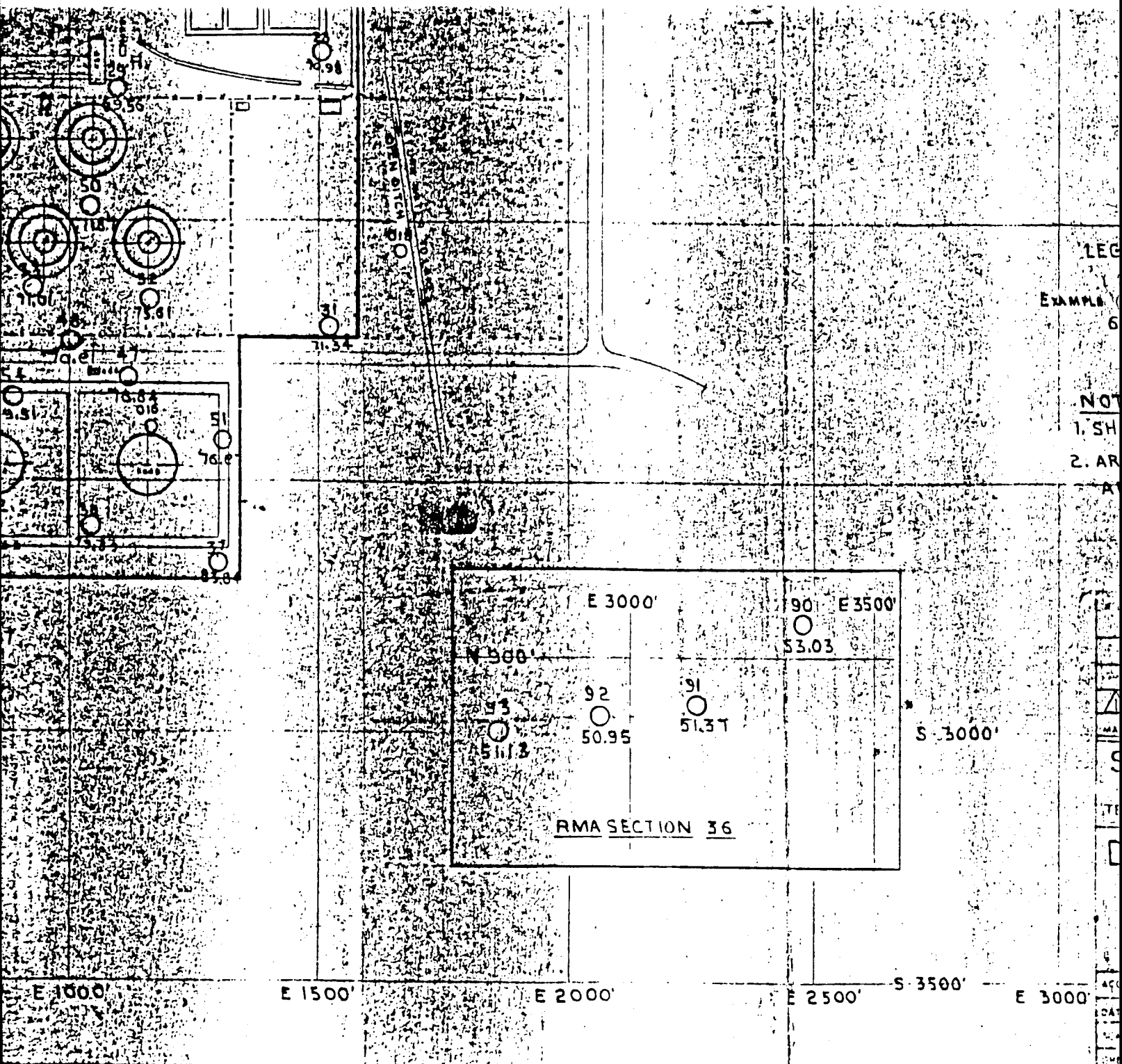
LEGEND

EXAMPLE 2-HOLE

66.2 ELEVATION TOP PIPE (+5200')

4068





EXAMPLE

6

Not

1. SH

2. AR

A

77

450

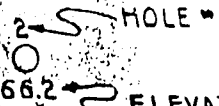
CAT

2008

7068

S 2000'

LEGEND

EXAMPLE 

NOTES

1. SHELL OBSERVATION HOLES ARE $\frac{3}{16}$ "
2. ARMY OBSERVATION HOLES ARE $\frac{1}{8}$ "

S 2500'

E 3000'

E 3500'

90
53.03

92
50.95

91
51.37

S 3000'

RMA SECTION 36

SHELL CHEMICAL COMPANY

A DIVISION OF SHELL OIL COMPANY

TECHNICAL DEPARTMENT

DENVER PLANT

DENVER CHEMICAL PLANT

83041R01
2nd Copy

ACCT NO 8042

DATE 9-7-79

SCALE 1" = 200' DEN DLH

CHECKED

APPROVED WDS

YE-13821-1

E 2000'

E 2500'

S 3500'

E 3000'

80/8